8/3/88

#9

CERTIFICATE OF TRANSLATION

AN FY OF TOKYO MUDASSY OF THE UNITED STATES OF AMERICA)
Charles E. Robertson, III Refore me, Consul
of the United States of America, in and for Tokyo, Japan,
duly commissioned and qualified, personally appeared
Kazuyo Saito , who, being duly sworn,
deposes and says:
That my name is Kazuyo Saito C/O KYOWA HAKKO KOGYO CO., LTD.
That my address is 6-1, Ohtemachi Itchome, Chiyoda-ku, Tokyo
That I know well both the English and Japanese
languages;
That I translated the attached Japanese language
document into the English Language;
That the attached English language translation is
a true and correct translation of the attached
Japanese language document to the best of my
knowledge and belief;
And further deponent saith not.
Kazuyo Saito

Charles E. Robertson, II
Consul of the
United States of America

SUBSCRIBED AND SUORH TO before me

this 22nd day of July , A.D. 1988

GCS-77 TKY 1/73

PATENT OFFICE JAPANESE GOVERNMENT

This is to certify that the annexed is a true copy of the following application as filed with this Office.

DATE OF APPLICATION: March 3, 1986

APPLICATION NUMBER:

45676/1986

APPLICANT:

KYOWA HAKKO KOGYO CO., LTD.

Dated this 13th day of July, 1988

Fumitake Yoshida Director-General Patent Office

Certified No. SHO 63-28306

APPLICATION FOR PATENT

March 3, 1986

To: Director-General Patent Office

- 1. Title of the Invention DIBENZ[b,e]OXEPIN DERIVATIVE AND ANTIALLERGIC AND ANTIINFLAMMATORY AGENT
- Number of Claims
- Inventors

Name : Etsuo Oshima (and other 8)

Address: Okazaki-mansion 305, 152-2, Sugihara, Nagaizumicho, Sunto-gun, Shizuoka-ken

4. Applicant

Name : (102) Kyowa Hakko Kogyo Co., Ltd.

Mikio Kato, Representative Director

Address: 6-1, Ohtemachi Itchome, Chiyoda-ku, Tokyo

Postal Code: 100

- 5. List of the Annexed Documents
 - (1) Specification

one copy

(2) Duplicate of the Application one copy

6. Inventors Other Than the Above

Name : Toshiaki Kumazawa

Address: 1194-83, Shimotogari, Nagaizumi-cho, Sunto-gun,

Shizuoka-ken

Name : Shizuo Otaki

Address: Seiwa-ryo, 1188, Shimotogari, Nagaizumi-cho,

Sunto-gun, Shizuoka-ken

Name : Hiroyuki Obase

Address: Kyowa-apart 204, 1188, Shimotogari, Nagaizumi-cho,

Sunto-gun, Shizuoka-ken

Name : Kenji Ohmori

Address: 2-14-3, Fuyodai, Mishima-shi, Shizuoka-ken

Name : Hidee Ishii

Address: 847-3, Nanjo, Nirayama-cho, Tagata-gun, Shizuoka-

ken

Name : Haruhiko Manabe

Address: Seiwa-ryo, 1188, Shimotogari, Nagaizumi-cho,

Sunto-gun, Shizuoka-ken

Name : Tadafumi Tamura

Address: Seiwa-ryo, 1188, Shimotogari, Nagaizumi-cho,

Sunto-gun, Shizuoka-ken

Name : Katsuichi Shuto

Address: 410-1, Nameri, Nagaizumi-cho, Sunto-gun, Shizuoka-

ken

SPECIFICATION

- - 2. Scope of Claim for Patent
- 10 (1) A dibenz[b,e]oxepin derivative represented by
 the formula (I)

$$\begin{array}{c}
X - (CH_2)_n - Z \\
\downarrow \\
Y - A
\end{array}$$
(I)

wherein

- A represents a hydroxymethyl, a lower alkoxymethyl, a

 triphenylmethyloxymethyl, a lower alkanoyloxymethyl,
 a lower alkanoyl, a carboxy, a lower alkoxy carbonyl,
 a triphenylmethyloxycarbonyl, -CONR₁R₂ (wherein R₁
 and R₂ are the same or different and represent a
 hydrogen atom or lower alkyl) , 4,4-dimethyl-2oxazoline-2-yl group or -CONHOH;
 - Y represents $-(CH_2)_m$, $-CHR_3$ - $(CH_2)_m$ or $-CR_4$ = CR_5 - $(CH_2)_m$ (wherein R_3 represents a lower alkyl, R_4 and R_5 are
 the same or different and represent a hydrogen atom
 or a lower alkyl, and m is 0, 1, 2, 3 or 4, which is
 the substituent at 2- or 3-position of the mother
 nucleus and the left side of the group Y is bound to
 benzen nucleus);
 - X represents = N-, = CH- or $-CH_2-$; n is 0, 1, 2, 3 or 4;
- Z represents 4-methylpiperazino, 4-methylhomopiperazino, piperidino, pyrrolidino, thiomorpholino, morpholino or $-NR_6R_7$ (wherein R_6 and R_7 are the same or different

and represent a hydrogen atom or a lower alkyl); and ____ means single bond or double bond; and the pharmaceutically acceptable salts thereof.

(2) An antiallergic agent containing, as an active ingredient, a dibenz[b,e]oxepin derivative represented by the formula

{wherein

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15 X represents = $N - = CH - or - CH_2 -$; n is 0, 1, 2, 3 or 4;

Z represents 4-methylpiperazino, 4-methylhomopiperazino, piperidino, pyrrolidino, thiomorpholino, morpholino or $-NR_6R_7$ (wherein R_6 and R_7 are the same or different and represent a hydrogen atom or a lower alkyl);

--- means single bond or double bond;

-Y'-A" represents, when X is = CH - or - CH_2 -, -Y-A [wherein

A represents a hydroxymethyl, a lower alkoxymethyl, a triphenylmethyloxymethyl, a lower alkanoyloxymethyl, a lower alkanoyl, a carboxy, a lower alkoxy carbonyl, a triphenylmethyloxycarbonyl, -CONR₁R₂ (wherein R₁ and R₂ are the same or different and represent a hydrogen atom or lower alkyl), 4,4-dimethyl-2-oxazoline-2-yl group or -CONHOH;

Y represents $-(CH_2)_m$, $-CHR_3$ - $(CH_2)_m$ - or $-CR_4$ = CR_5 - $(CH_2)_m$ - (wherein R_3 represents a lower alkyl, R_4 and R_5 are the same or different and represent a hydrogen atom or a lower alkyl, and m is 0, 1, 2, 3 or 4, which is the substituent at 2-or 3-position of the mother nucleus and the left side of the group Y is bound to benzen nucleus)];

or represents, when X is = N-, -Y-A which is bound at 2-position of the mother nucleus (wherein Y and A have the same meanings as previously defined)}

or the pharmaceutically acceptable salts thereof.

(3) An antiinflammatory agent containing, as an active ingredient, a dibenz[b,e]oxepin derivative represented by the formula

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[wherein

n is 0, 1, 2, 3 or 4;

Z represents 4-methylpiperazino, 4-methylhomopiperazino, piperidino, pyrrolidino, thiomorpholino, morpholino or $-NR_6R_7$ (wherein R_6 and R_7 are the same or different and represent a hydrogen atom or a lower alkyl);

Y" represents $-CH_2$ or $-CHR_3$

(wherein R_3 represents a lower alkyl) which is the substituent at 2- or 3-position of the mother nucleus;

A'" represents a hydroxymethyl, a lower alkoxymethyl, a triphenylmethyloxymethyl, a lower alkanoyloxymethyl, a formyl, a carboxy, a lower alkoxy carbonyl, a triphenylmethyloxycarbonyl, -CONR₁R₂ (wherein R₁ and R₂ are the same or different and represent a hydrogen atom or lower alkyl], 4,4-dimethyl-2-oxazoline-2-yl group or -CONHOH;

or the pharmaceutically acceptable salts thereof.

3. Detailed Description of the Invention

Industrially Applicable Field

The present invention relates to a novel dibenz [b,e]oxepin derivative and to an antiallergic and/or anti-inflammatory agent containing the same as an active ingredient.

Prior Art

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Heretofore, it has been known that ll-unsubstituted, ll-hydroxy or ll-oxodibenz[b,e]oxepin derivative
is used for antiinflammatory agents [J. Med. Chem., 21,
633-639 (1978)].

Further, it is known that dibenz[b,e]oxepin derivative wherein substituents Ra and Rb at 11-position have the following definitions, is employed in the treatment and control of allergic conditions (USP 4,282,365).

Ra : H, OH, lower alkoxy, lower alkylthio, lower alkylsulfinyl, lower alkylsulfonyl, arylthio, NH2, NHCHO or imidazolyl;

Rb : H or lower alkyl;

or Ra and Rb taken together are = 0, =CH-Rc (wherein Rc is H or aryl).

Furthermore, it is known that ll-(4-methyl-piperazino) dibenz[b,e]oxepin derivative has an antiasthmatic activity (JP-A-150082/81).

It is also known that dibenz[b,e]oxepin derivative having the following formula:

35 (wherein Rd and Re are lower alkyl and Rf is lower alkyl or halogen) has an antiasthmatic activity (JP-A-126883/83).

Dibenz[b,e]oxepin derivative having an antiallergic activity and having the following structural formula:

10 (wherein Rg and Rh are alkyl, r is 2 or 3 and Ri is alkyl or halogen) is known (JP-A-227879/84).

Dibenz[b,e]oxepin derivative having an antiallergic activity and having the following structural formula:

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[wherein R_j is 4-alkylpiperazino, 3-quinuclidylamino or $-Xa-(CH_2)_s$ -NR $_\ell$ R $_m$ (wherein X_a is -NH-, -S- or -O-, s is 2 or 3 and R_ℓ and R_m are alkyl), and R_k is CN, 5-tetrazolyl, $CONH_2$ or CO_2R_n (wherein R_n is H, alkyl or 1-(ethoxycarbonyloxy)ethyl)] is known (JP-A-28972/85).

Doxepin having an antidepressant activity and having the following structural formula is known [Drugs, 13, 161 (1977)].

Dothiepin having an antidepressant activity and having the following structural formula is known [Arz.-Forsch., 13 1039 (1963); ibid., 14 100 (1964)].

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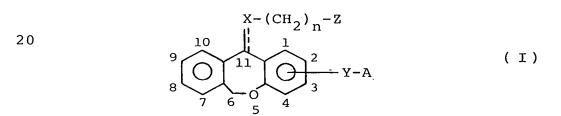
As the compound having both an antiallergic activity and an antiinflammatory activity, antiinflammatory steroids are known.

Problems to be Solved by the Invention

It is always desired that a novel and useful compound having an antiallergic activity or an anti-inflammatory activity be developed.

15 Means for Solving Problems

The repsent invention relates to a dibenz[b,e] oxepin derivative represented by the fomrula (I):



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[wherein A represents a hydroxymethyl, a lower alkoxymethyl, a triphenylmethyloxymethyl, a lower alkanoyloxymethyl, a lower alkanoyloxymethyl, a lower alkanoyl, a carboxy, a lower alkoxy carbonyl, a triphenylmethyloxycarbonyl, $-\text{CONR}_1\text{R}_2$ (wherein R_1 and R_2 are the same or different and represent a hydrogen atom or lower alkyl) 4,4-dimethyl-2-oxazoline-2-yl group or -CONHOH; Y represents $-(\text{CH}_2)_m$, $-\text{CHR}_3$ -(CH $_2$) $_m$ - or $-\text{CR}_4$ = CR $_5$ -(CH $_2$) $_m$ - which is substituent at 2- or 3-position of the mother nucleus (wherein R_3 represents a lower alkyl, R_4 and R_5 are the same or different and represent a hydrogen atom or a lower alkyl, m is 0, 1, 2, 3 or 4, and the left side of the group of Y mentioned above is bound to benzen nucleus); X represents

= N-, =CH- or -CH $_2$ -; n is 0, 1, 2, 3 or 4; Z represents 4-methylpiperazino, 4-methylhomopiperazino, piperidino, pyrrolidino, thiomorpholino, morpholino, or -NR $_6$ R $_7$ (wherein R $_6$ and R $_7$ are the same or different and represent a hydrogen atom or a lower alkyl); and ____ means a single bond or double bond] [hereinafter referred to as Compound (I) and Compounds with other formula numbers are hereinafter likewise referred to], and a pharmaceutically acceptable salt thereof. The present invention further pertains to an antiallergic or antiinflammatory agent containing at least one of Compound (I) and a pharmaceutically acceptable salt thereof as an active ingredient.

In the definition of each group of formula (I), the lower alkyl group includes straight or branched chain alkyl groups having 1 to 6 carbon atoms, for example, methyl, ethyl, n-propyl, iso-propyl, n-butyl, etc.

In the definition of the group A, lower alkyl moiety of lower alkoxymethyl group and lower alkoxycarbonyl group has the same meaning as previously defined.

The lower alkoxymethyl group includes methoxymethyl, ethoxymethyl, n-propoxymethyl, isopropoxy, etc. and the lower alkoxycarbonyl group includes methoxycarbonyl, etc.

In the definition of the group A, the lower alkyl moiety of lower alkanoyl group and lower alkanoyloxy-methyl group has the same meaning as previously defined.

The lower alkanoyl group includes formyl, acetyl, etc. and the lower alkanoyloxymethyl group includes formyloxymethyl, acetyloxymethyl, etc.

The pharmaceutically acceptable salt of Compound (I) includes pharmaceutically acceptable acid addition salt, metal salt, ammonium salt, organic amine addition salt, amino acid addition salt, etc.

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The pharmaceutically acceptable acid addition salt of Compound (I) includes inorganic acid salts such as hydrochloride, sulfate, phosphate, etc., and organic acid salts such as acetate, maleate, fumarate, tartrate, citrate, etc. The pharmaceutically acceptable metal salt includes alkalimetal salts such as sodium salt, potassium salt, etc., alkaline earch metal salts such as magnesium salt, calcium salt, etc., and alminium salt, zinc salt, etc. The pharmaceutically acceptable organic amine addition salt includes addition salt of morpholine and piperidine and the pharmaceutically acceptable amino acid addition salt includes addition salt of lysine, glycine, phenylalanine, etc.

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Compound (I) is prepared by using a compound 15 represented by the formula (II):

(wherein Y and A have the same meanings as previously defined) or a compound represented by the formula (III):

(wherein Y and A have the same meanings as previously defined) as the starting compound. Compound (II) is disclosed in J. Med. Chem., 19, 941 (1976), ibid., 20, 1499 (1977) and JP-A-21679/83.

Compound (III) wherein -Y-A is -COOH is dis
closed in JP-A-21679/83 and the other Compounds (III) can
be prepared according to the method described in the
publication though they do not occur in the publication.

The process for preparing Compound (I) is explained, depending on the kind of the group X.

Process A

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5 [Synthesis of Compound (I) wherein X is =CH- (Part 1)]

The carboxy group of Compound (IIa) is protected according to the following reaction scheme.

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$$\xrightarrow{\text{SOCl}_2}$$
 $\xrightarrow{\text{SOCl}_2}$ $\xrightarrow{\text{CH}_3}$ $\xrightarrow{\text{CH}_3}$ $\xrightarrow{\text{CH}_3}$

$$\begin{array}{c}
 & \text{SOCl}_2 \\
 & \text{(V)}
\end{array}$$

In the formulae, Y has the same meaning as previously defined, and Compound (IIa) is included in Compound (II) (compounds with an alphabet suffix following formula number are likewise included in compounds with common formula no.).

Compound (IIa) is reacted with 1 to 5 equivalents of thionyl chloride and 1 to 5 equivalents of 2-amino-2-methyl-1-propanol on the basis of Compound (IIa) in an inert solvent such as methylene chloride, if necessary in the presence of a base such as triethylamine at a temperature of from 0°C to room temperature for 1 - 24 hours to form Compound (IV). Compound (IV) can also be obtained by reacting Compound (IIa) with thionyl chloride in advance and then with 2-amino-2-methyl-1-propanol.

Compound (IV) is reacted with 1-5 equivalents of thionyl chloride in an inert solvent such as methylene chloride, toluene and benzene at a temperature

of from 0°C to room temperature for 1-24 hours to form Compound (V).

Compounds (Ia) and (Ib) can be prepared from Compound (V) according to the following reaction scheme.

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HO_{$$T_{2}$$}

(CH₂) n+1 Z

(CH₂) n+1 Z

(CH₂) n Z

(CH₂) n Z

(VII)

(Ia)

R₈'OH

(CH₂) n Z

In the formulae, Y, Z, and n have the same menaings as previously defined, R₈ is hydrogen or a lower alkyl group, R'₈ is a lower alkyl group and Hal is halogen.

As used herein, the term lower alkyl has the same meaning as that of lower alkyl in each group of formula (I). Halogen includes chlorine, bromine and iodine.

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Compound (V) is reacted with 1-5 equivalents of Compound (VI) in an inert solvent such as tetrahydrofuran and diethyl ether under atmosphere of an inert gas such as nitrogen and argon to form Compound (VII). The reaction is carried out at a temperature of from 0°C to room temperature and is usually completed in 1-24 hours.

Compound (VII) is reacted with 1-5 equivalents of thionyl chloride or phosphoryl chloride in an inert solvent such as methylene chloride in the presence of a base such as pyridine to form Compound (Ia). The reaction is carried out at a temperature of from 0°C to room temperature and is completed in 1-24 hours.

Compound (Ia) is allowed to stand in an alcohol containing water, such as aqueous methanol solution, in the presence of an appropriate acidic catalyst such as ptoluenesulfonic acid at a temperature of from room temperature to the boiling point of the solvent to form Compound (Ib) wherein R_8 is H. The reaction is completed in 1-24 hours.

Compound (VII) is allowed to stand in an alcohol of R_8 'OH in the presence of an appropriate acidic catalyst such as p-toluenesulfonic acid at a temperature of from room temperature to the boiling point of the solvent to form Compound (Ib) wherein R_8 is a lower alkyl. The reaction is completed in 1-24 hours.

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Process B

[Synthesis of Compound (I) wherein X is =CH- (Part 2)]

The carboxy group of a compound represented by
the formula (IIa) can be converted to a lower alkoxymethyl
group or a trityloxymethyl group according to the
following reaction scheme.

In the formulae, Y has the same meaning as previously defined, R_9 is a lower alkyl group and R_9 ' is a trityl group or a lower alkyl group. The term lower alkyl has the same meaning as that of lower alkyl in each group in formula (I).

Compound (IIa) is reduced with 1-5 equivalents of lithium aluminium hydride in tetrahydrofuran at a temperature of from 0°C to room temperature for 1-24 hours to form Compound (VIII).

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Compound (VIII) is reacted with $1-5\ \text{equivalents}$ of trityl chloride in pyridine at a temperature

of from room temperature to 100° C for 1-24 hours to form Compound (IX).

Compound (IX) is oxidized with 1-5 equivalents of an appropriate oxidizing agent such as potassium permanganate and pyridinium-chlorochromate reagent in an inert solvent such as methylene chloride and acetone to form Compound (XI) wherein R_9 ' is trityl. The reaction is carried out at a temperature of from 0°C to the boiling point of the solvent and is completed in 1-24 hours.

Compound (VIII) is allowed to stand in an alcohol of $R_9^{\,\rm OH}$ in the presence of an appropriate acidic catalyst such as sulfuric acid at a temperature of from room temperature to the boiling point of the solvent to form Compound (X). The reaction is usually completed in 1-24 hours.

Compound (X) is oxidized with 1-5 equivalents of an appropriate oxidizing agent such as Jones reagent in an inert solvent such as acetone to form Compound (XI) wherein R_9 ' is a lower alkyl. The reaction is carried out at a temperature of from 0°C to the boiling point of the solvent and is usually completed in 1-24 hours.

The compounds represented by the formulae (Ic) and (Id) and if desired, the compound represented by the formula (Ie) can be synthesized from Compound (XI) according to the following reaction scheme.

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In the formulae, Y, Z, R_9 ', n and Hal have the same meanings as previously defined.

Compound (XI) is reacted with Compound (VI)

30 which is Grignard reagent according to the same manner as
in the reaction step from Compound (V) to Compound (VII)
in Process A to form Compound (XII).

Compound (XII) is subjected to reaction according to the same manner as in the reaction step from Compound (VII) to Compound (Ia) in Process A to form Compound (Ic).

Compound (Ic) is allowed to stand in a solvent containing water such as aqueous dioxane in the presence of an appropriate acidic catalyst such as p-toluenesulfonic acid at a temperature of from room temperature to the boiling point of the solvent to form Compound (Id). The reaction is usually completed in 1-24 hours.

Compound (Id) can also be obtained in one step by being allowed to stand Compound (XII) in a solvent containing water such as aqueous dioxane in the presence of an appropriate acidic catalyst such as sulfonic acid at a temperature of from room temperature to the boiling point of the solvent. The reaction is usually completed in 1-24 hours.

If desired, Compound (Id) is oxidized with 1-5 equivalents of an appropriate oxidizing agent such as Jones reagent in an inert solvent such as acetone to form Compound (Ie). The reaction is carried out at a temperature of from 0°C to the boiling point of the solvent and is usually completed in 1-24 hours.

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Process C

[Synthesis of Compound (I) wherein X is =CH- (Part 3)]

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$$\begin{array}{c} O \\ V-A' + Ph_3P = CH(CH_2)_nZ \\ (XIII) \end{array}$$

In the formulae, Y, Z, and n have the same meanings as previously defined. A' represents the groups falling within the definition of A but lower alkanoyl group.

Compound (IIb) is reacted with 1-5 equivalents of Compound (XIII) in an inert solvent such as tetrahydrofuran under atmosphere of an inert gas such as nitrogen and argon at a temperature of from 0°C to room temperature for 1-24 hours to form Compound (If).

Compound (XIII) which is ylide, can be prepared according to the method described in C.A. 63 16366a (1965).

$$Ph_3\underline{P} + Hal(CH_2)_{n+1}Hal \longrightarrow P\underline{h}_3P(CH_2)_{n+1}Hal \cdot Hal$$
(XIV)
(XV)

$$\frac{1) \quad HZ}{2) \quad HHal} \rightarrow Ph_3^P(CH_2)_{n+1}^Z \cdot Hal^- \cdot (HHal)_q$$
(XVI)

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In the formulae, Hal, n and Z have the same meanings as previously defined and q is 1 or 2.

Compound (XIV) is reacted with an equivalent of triphenylphosphine in toluene at reflux of the solvent for 1-24 hours to form Compound (XV).

Compound (XV) is reacted with 1-5 equivalents of HZ in ethanol at reflux of the solvent for 1-24 hours and excess HZ is distilled away under reduced pressure. After the addition of 1-5 equivalents of HHal on the basis of Compound (XV), the mixture is allowed to stand at a temperature of from 0°C to the boiling point of the solvent for 1-24 hours to form Compound (XVI) which is Wittig reagent.

of an appropriate base such as n-butyl lithium in an inert solvent such as tetrahydrofuran under atmosphere of an inert gas such as nitrogen and argon to form ylide (XIII). The reaction is carried out at -78°C ~ room temperature and is usually completed in 1-24 hours.

Process D

[Synthesis of Compound (I) wherein X is =CH- (Part 4)]

In the formulae, Y, Z and A have the same meanings as previously defined.

The process is known as Prins reaction [New Experimental Chemical Course (Maruzen), Vol. 14, Synthesis and Reaction of Organic Compound III, page 1375 (1977)].

Compound (III), 1 to 5 equivalents of

formaldehyde and 1 to 5 equivalents of HZ are
subjected to reaction in an inert solvent such as tetrachloroethane in the presence of an acid or reaction in
an acid as such serving as a solvent under atmosphere of
an inert gas such as nitrogen and argon to yield Compound

(Ig).

The formaldehyde or polymerized formaldehyde includes p-formaldehyde, trioxane, etc. The acid includes acetic acid, trichloroacetic acid, trifluoroacetic acid, etc. The reaction is carried out at a temperature of from room temperature to the boiling point of the solvent and is completed in 1-24 hours.

Compound (III) which is the starting material can be prepared according to the process described in JP-A-21679/83, as shown below.

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That is, Compound (IIb), 1 to 5 equivalents of methyltriphenylphosphonium bromide and 1 to 5 equivalents of n-butyl lithium on the basis of

Compound (IIb) are subjected to reaction in an inert solvent at from -78°C to room temperature for 1 to 5 hours to yield ylide (XVII) which is reacted with an equivalents of Compound (IIb) in an inert solvent at from -78°C to room temperature under atmosphere of an inert gas for 1 to 24 hours to yield Compould (IIIa).

The inert gas includes nitrogen, argon, etc. and the inert solvent includes tetrahydrofuran, etc.

The group A' in Compound (IIIa) can easily be converted to a lower alkanoyl group as is stated in Process I and therefore, Compound (III) can easily be prepared.

Process E

[Synthesis of Compound (I) wherein X is=N-]

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Y-A' +
$$H_2N(CH_2)_nZ$$

(XVIII)

N $\sim (CH_2)_nZ$

(Ih)

Compound (IIb) and 1 to 10 equivalents of Compound (XVIII) are subjected to reaction in an inert solvent such as benzene in the presence of 1 to 10 equivalents of titanium tetrachloride at from 0°C to the boiling point of the solvent under atmosphere of an inert gas such as nitrogen and argon for 1 to 48 hours to yield Compound (Ih).

Process F

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10 [Synthesis of Compound (I) wherein X is -CH₂- (Part 1)]

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Chlorination

(XIX)

Chlorination

(XIX)

Chlorination

(XIX)

Chlorination

(XIX)

$$(XIX)$$
 (XIX)
 $(XI$

In the formulae, Y, Z, n, R_8 and Hal have the same meanings as previously defined.

Compound (V) is reduced with 1 to 5 equivalents of lithium aluminium hydride or sodium borohydride in an inert solvent such as tetrahydrofuran and methanol at from 0°C to room temperature for 1 to 24 hours to yield Compound (XIX).

Compound (XIX) and 1 to 5 equivalents of thionyl chloride or phosphoryl chloride are subjected to reaction in an appropriate base such as pyridine at from

0°C to room temperature to yield Compound (XX).

Compound (XX) and 1 to 5 equivalents of Compound (VI) are subjected to reaction in the same manner as in the reaction step from Compound (V) to Compound (VII) in Process A to yield Compound (Ii).

Compound (Ii) is subjected to reaction in the same manner as in the reaction step from Compound (VII) to Compound (Ib) or the reaction step from Compound (Ia) to Compound (Ib) in Process A to yield Compound (Ij).

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Process G

[Synthesis of Compound (I) wherein X is -CH2- (Part 2)]

$$\begin{array}{c}
\text{HalMg}(CH_2)_{n+1}Z \text{ (VI)} \\
& \longrightarrow \\
\text{(Ik)}
\end{array}$$

$$\begin{array}{c}
\text{CH}_2 - (CH_2)_n Z \\
\text{Y-CH}_2 \text{OR}_9
\end{array}$$

Compound (XXI) is subjected to chlorination in the same manner as in Process F to yield Compound (XXII).

Compound (XXII) and Compound (VI) are subjected to reaction in the same manner as in Process F to yield Compound (Ik).

Compound (Ik) is treated in the same manner as in Process B to form Compound (Il).

Compound (I1) is further treated to form Compound (Im).

Compound (IX) is included in the definition of the starting material (XXI).

Compound (XI) is reduced with 1 to 5 equivalents of lithium alminium hydride or sodium borohydride in an inert solvent such as tetrahydrofuran and methanol at from 0°C to room temperature for 1 to 24 hours to yield Compound (XXI).

Process H

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[Synthesis of Compound (I) wherein X is -CH₂- (Part 3)]

Synthesis of Compound (I) wherein X is -CH₂- is illustrated in Processes F and G, but is not restricted thereto. For example, compound (I) wherein X is -CH₂- can also be prepared by subjecting Compounds (Ia) - (Ig)

obtained by Processes A - D to appropriate reduction such as hydrogenation using palladium-carbon as catalyst.

Process I

In processes A to H, there are cases where compounds having the group defined as A are not directly obtained, for example, Compound (I) wherein A is a lower alkanoyl group is not obtained in Process C. However, compounds having the group defined as A can easily be synthesized by subjecting the products obtained by each process to an ordinary procedure in organic chemical synthesis such as oxidation, reduction and hydrolyzation. For example, Compound (IIb), the starting material having a corresponding hydroxymethyl group is treated according to Process C and then an oxidative reaction is carried out to obtain Compound (I) wherein A is a formyl group.

The intermediates and the desired compounds in each of the processes described above can be purified and isolated by a purification method which is usually used in the field of organic chemical synthesis, such as filtration, extraction with organic solvent such as ethyl acetate and methylene chloride, drying, concentration, recrystallization, column chromatography, etc.

Out of Compounds (Ia) - (Ih) obtained in each of the processes described above, with regard to stereochemistry at ll-position of dibenz[b,e]oxepin, Compounds (Ia), (Ib), (Ic), (Id), (Ig) and (Ih) are apt to be formed as a trans-form and Compound (If) is apt to be formed as a cis-form, with high frequency compared with the other form.

When Compound (I) except Compounds (Ii) - (Im) is produced as a cis-trans mixture, Compound (I) is separated and purified by an appropriate method which is usually used in the field of organic chemical synthesis, such as column chromatography, recrystallization, etc.

If desired, cis-form can be converted to transform. For example, cis-form is added to an acetic acid and the mixture is heated at reflux in the presence of an appropriate catalyst such as p-toluenesulfonic acid for 1-24 hours to form trans-form.

With regard to the denotation of cis-form (or syn-form) and trans-form (or anti-form) of Compound (I), Compound (I) wherein the substituent bound to the double bond is on the same side as oxygen of oxepin, is cis-form (or syn-form) and Compound (I) wherein the substituent is on the opposite side is trans-form (or anti-form).

For example, the compound represented by the following formula is cis-form.

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When it is desired to obtain a salt of Compound (I), in case of obtaining Compound (I) in the form of salt, the salt may be purified as such, or in case of obtaining Compound (I) in a free form, a salt thereof may be formed according to an ordinary procedure.

Table 1 shows examples of Compound (I) or pharmaceutically acceptable salts thereof and Table 2 shows the structural formula thereof.

Table 3 shows characteristic signals in NMR and Table 4 shows retention time in HPLC.

Table 1

	Compound	
	No.	Compound (I)
5	1	Methyl cis-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
	_	Methyl trans-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
10	2	Ethyl cis-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
	-	Ethyl trans-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
15	3	Cis-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic acid
)	Trans-ll-(3-dimethylaminopropylidene)-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid
	4	Methyl cis-ll-(3-diethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
20	-	Methyl trans-ll-(3-diethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
	. 5	Cis-ll-(3-diethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic acid
25	3	Trans-11-(3-diethylaminopropylidene)-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid
	6	Methyl cis-ll-(3-pyrrolidinopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
30	Ŭ	Methyl trans-ll-(3-pyrrolidinopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
	7	Cis-ll-(3-pyrrolidinopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic acid
35	,	Trans-ll-(3-pyrrolidinopropylidene)-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid
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5	8	Methyl cis-ll-(4-dimethylaminobutylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate Methyl trans-ll-(4-dimethylaminobutylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
	9	Cis-ll-(4-dimethylaminobutylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic acid Trans-ll-(4-dimethylaminobutylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic acid
10	10	Methyl cis-ll-[2-(4-methylpiperazino)- ethylidene]-6,ll-dihydrodibenz[b,e]oxepin-2- carboxylate Methyl trans-ll-[2-(4-methylpiperazino)- ethylidene]-6,ll-dihydrodibenz[b,e]oxepin-2- carboxylate
•	11	Cis-11-[2-(4-methylpiperazino)ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid Trans-11-[2-(4-methylpiperazino)ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid
20	12	Methyl cis-ll-(2-morpholinoethylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate Methyl trans-ll-(2-morpholinoethylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
25	13	Cis-ll-(2-morpholinoethylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic acid Trans-ll-(2-morpholinoethylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic acid
30	14	Methyl cis-ll-(2-thiomorpholinoethylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate Methyl trans-ll-(2-thiomorpholinoethylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
35	15	Cis-ll-(2-thiomorpholinoethylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic acid Trans-ll-(2-thiomorpholinoethylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic acid

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5	16	Methyl cis-ll-(2-pyrrolidinoethylidene)-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate Methyl trans-ll-(2-pyrrolidinoethylidene)-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate
	17	Methyl cis-ll-(2-piperidinoethylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate Methyl trans-ll-(2-piperidinoethylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
10	18	Methyl cis-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-acetate Methyl trans-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-acetate
15	19	Ethyl cis-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-acetate Ethyl trans-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-acetate
20	20	Cis-11-(3-dimethylaminopropylidene)-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid Trans-11-(3-dimethylaminopropylidene)-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid
25	21	Methyl cis-ll-(4-dimethylaminobutylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-acetate Methyl trans-ll-(4-dimethylaminobutylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-acetate
30	22	Cis-ll-(4-dimethylaminobutylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid Trans-ll-(4-dimethylaminobutylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid
35	23	Methyl cis-ll-(3-pyrrolidinopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-acetate Methyl trans-ll-(3-pyrrolidinopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-acetate

	24	Cis-ll-(3-pyrrolidinopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid
5	_ ·	Trans-11-(3-pyrrolidinopropylidene)-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid
	25	Methyl cis-11-[2-(4-methylpiperazino)- ethylidene-6,11-dihydrodibenz[b,e]oxepin-2- acetate
10	25	Methyl trans-ll-[2-(4-methylpiperazino)- ethylidene-6,ll-dihydrodibenz[b,e]oxepin-2- acetate
	26	Cis-ll-[2-(4-methylpiperazino)-ethylidene- 6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid
15		Trans-11-[2-(4-methylpiperazino)-ethylidene-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid
,	27	Methyl cis-3-[ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-yl]-propionate
20	2,	Methyl trans-3-[11-(3-dimethylaminopropyli-dene)-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionate
	28	Cis-3-[11-(3-dimethylaminopropylidene)-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid
		Trans-3-[11-(3-dimethylaminopropylidene)-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid
25	29	Methyl cis-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-3-acetate
		Methyl trans-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-3-acetate
30	30	Cis-ll-(3-dimethylaminopropylidene)-6,11-dihydrodibenz[b,e]oxepin-3-acetic acid
		Trans-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-3-acetic acid
35	31	Cis-ll-(3-dimethylaminopropylidene)-2-(2-hydroxyethyl)-6,ll-dihydrodibenz[b,e]oxepin
		Trans-11-(3-dimethylaminopropylidene)-2-(2-hydroxyethyl)-6,ll-dihydrodibenz[b,e]oxepin
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5	32	Cis-11-(3-dimethylaminopropylidene)-2-(2-triphenylmethyloxymethyl)-6,11-dihydrodibenz-[b,e]oxepin Trans-11-(3-dimethylaminopropylidene)-2-(2-triphenylmethyloxymethyl)-6,11-dihydrodibenz-[b,e]oxepin
10	33	Cis-ll-(3-dimethylaminopropylidene)-2-(3-hydroxypropyl)-6,ll-dihydrodibenz[b,e]oxepin Trans-ll-(3-dimethylaminopropylidene)-2-(3-hydroxypropyl)-6,ll-dihydrodibenz[b,e]oxepin
15	34	Methyl syn-ll-(2-diethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate Methyl anti-ll-(2-diethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate
:	35	Syn-ll-(2-diethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic acid Anti-ll-(2-diethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic acid
20	36	Methyl syn-ll-(2-dimethylaminoethyl)imino- 6,ll-dihydrodibenz[b,e]oxepin-2-acetate Methyl anti-ll-(2-dimethylaminoethyl)imino- 6,ll-dihydrodibenz[b,e]oxepin-2-acetate
25	37	Syn-ll-(2-dimethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid Anti-ll-(2-dimethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid
30	38	Methyl syn-ll-(2-diethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-acetate Methyl anti-ll-(2-diethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-acetate
35 39		Syn-11-(2-diethylaminoethyl)imino-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid Anti-11-(2-diethylaminoethyl)imino-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid
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5	40	Methyl syn-ll-(3-dimethylaminopropyl)imino- 6,ll-dihydrodibenz[b,e]oxepin-2-acetate Methyl anti-ll-(3-dimethylaminopropyl)imino- 6,ll-dihydrodibenz[b,e]oxepin-2-acetate
	41	Syn-ll-(3-dimethylaminopropyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid
10		Anti-11-(3-dimethylaminopropyl)imino-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid
10	42	Methyl syn-3-[11-(2-diethylaminoethyl)imino- 6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionate
:		Methyl anti-3-[ll-(2-diethylaminoethyl)imino- 6,ll-dihydrodibenz[b,e]oxepin-2-yl]-propionate
15	43	Syn-[11-(2-diethylaminoethyl)imino-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid
		Anti-[11-(2-diethylaminoethyl)imino-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid
20	44	Methyl syn-2-[11-(2-dimethylaminoethyl)imino- 6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionate
		Methyl anti-2-[ll-(2-dimethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-yl]-propionate
25	45	Syn-2-[11-(2-dimethylaminoethyl)imino-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid
		Anti-2-[ll-(2-dimethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid
-	46	Methyl syn-ll-(2-dimethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-3-acetate
30	40	Methyl anti-ll-(2-dimethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-3-acetate
٠.	47	Syn-11-(2-dimethylaminoethyl)imino-6,11-dihydrodibenz[b,e]oxepin-3-acetic acid
35	* /	Anti-11-(2-dimethylaminoethyl)imino-6,11-dihydrodibenz[b,e]oxepin-3-acetic acid
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Methyl syn-11-(3-dimethylaminopropyl)imino-6,11-dihydrodibenz[b,e]oxepin-3-acetate Methyl anti-11-(3-dimethylaminopropyl)imino-6,11-dihydrodibenz[b,e]oxepin-3-acetate Syn-11-(3-dimethylaminopropyl)imino-6,11-dihydrodibenz[b,e]oxepin-3-acetic acid Anti-11-(3-dimethylaminopropyl)imino-6,11-dihydrodibenz[b,e]oxepin-3-acetic acid 10 50 Methyl 11-(3-dimethylaminopropyl)-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate 51 11-(3-dimethylaminopropyl)-6,11-dihydrodibenz-[b,e]oxepin-2-acetic acid 52 11-(3-dimethylaminopropyl)-6,11-dihydrodibenz-[b,e]oxepin-2-acetic acid 53 11-(3-Dimethylaminopropyl)-6,11-dihydrodibenz-[b,e]oxepin 54 11-(3-Dimethylaminopropyl)-2-(4,4-dimethyl-2-oxazoline-2-yl)-6,11-dihydrodibenz[b,e]oxepin			<u> </u>	
Syn-ll-(3-dimethylaminopropyl)imino-6,ll-dihydrodibenz[b,e]oxepin-3-acetic acid Anti-ll-(3-dimethylaminopropyl)imino-6,ll-dihydrodibenz[b,e]oxepin-3-acetic acid Methyl ll-(3-dimethylaminopropyl)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate 51	5	48	Methyl anti-11-(3-dimethylaminopropyl)imino-	
dinydrodibenz[b,e]oxepin-3-acetic acid Methyl ll-(3-dimethylaminopropyl)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate 11-(3-dimethylaminopropyl)-6,ll-dihydrodibenz-[b,e]oxepin-2-carboxylic acid 12	3	49	Syn-ll-(3-dimethylaminopropyl)imino-6,ll-dihydrodibenz[b,e]oxepin-3-acetic acid Anti-ll-(3-dimethylaminopropyl)imino-6,ll-	
15 52 11-(3-dimethylaminopropyl)-6,11-dihydrodibenz- [b,e]oxepin-2-acetic acid 11-(3-Dimethylaminopropylidene)-2-(4,4- dimethyl-2-oxazoline-2-yl)-6,11-dihydrodibenz- [b,e]oxepin 11-(3-Dimethylaminopropyl)-2-(4,4-dimethyl-2-	10	50	Methyl 11-(3-dimethylaminopropyl)-6 11-	
11-(3-Dimethylaminopropylidene)-2-(4,4-dimethyl-2-oxazoline-2-yl)-6,11-dihydrodibenz- [b,e]oxepin 11-(3-Dimethylaminopropyl)-2-(4,4-dimethyl-2-	15	51	11-(3-dimethylaminopropyl)-6,11-dihydrodibenz- [b,e]oxepin-2-carboxylic acid	
20 dimethyl-2-oxazoline-2-yl)-6,ll-dihydrodibenz- [b,e]oxepin 11-(3-Dimethylaminopropyl)-2-(4,4-dimethyl-2-		. 52	11-(3-dimethylaminopropyl)-6,11-dihydrodibenz- [b,e]oxepin-2-acetic acid	
11-(3-Dimethylaminopropyl)-2-(4,4-dimethyl-2-oxazoline-2-yl)-6,ll-dihydrodibenz[b,e]oxepin	20	53	$a_{\text{out}} = a_{\text{out}} = a_{$	
		54	<pre>11-(3-Dimethylaminopropyl)-2-(4,4-dimethyl-2- oxazoline-2-yl)-6,11-dihydrodibenz[b,e]oxepin</pre>	

	Compound No.	
5	3'	1/2 Fumarate · 1/5 hydrate of Compound 3 (trans form 99%)
	5 '	Fumarate · 1/3 hydrate of Compound 5 (cis form 99%)
10	7'	Fumarate · 1 hydrate of Compound 7 (cis form 70%)
	11'	2 Fumarate · 1/2 hydrate of Compound 11 (trans form 100%)
	13'	1/2 Fumarate ·1/2 hydrate of Compound 13 (trans form 93%)
15	15'	Fumarate of Compound 15 (trans form 100%)
	20'	Fumarate · 3/2 hydrate of Compound 20 (trans form 95%)
	26'	Fumarate · 2/3 hydrate of Compound 26 (trans form 88%)
20	28'	Fumarate · 1/2 hydrate of Compound 28 (trans form 63%)
	31'	1/2 Fumarate · 1 hydrate of Compound 31 (trans form 95%)
25	33'	Fumarate of Compound 33 (cis form 100%)
	35'	Sodium salt · 1 hydrate of Compound 35 (anti:syn = 1:1)
	43'	Sodium salt of Compound 43 (anti form 98%)
30	. 45 '	Sodium salt · 1 hydrate of Compound 45 (anti form 99%)
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Table 2

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 $X-(CH_2)_n-Z$ Me: methyl group Ph: phenyl group Et: ethyl group

	Compound No.	X	-Y-A	-(CH ₂) _n -z
10	1	СН	2-COOMe	→ NMe ₂
	2	11	2-COOEt	n ·
	3	u ·	2-СООН	п
15	4	n	2-COOMe	NEt ₂
	5	11	2-СООН	n
	6	11	2-COOMe	\sim N
20	7	11	2-СООН	11
	8	11	2-COOMe	NMe ₂
	9		2-СООН	"
25	10	11	2-COOMe	NMe
	11	11	2-СООН	и
	12	ır	2-COOMe	N O
30	13	. 11	2-СООН	u ·
	14	tt .	2-COOMe	N S
	15	.11	2-СООН	tt .
35	16	11	2-C00Me	N
	17	11	2-COOMe	N

	Compound No.	х	-Y-A	-(CH ₂) _n -z
5	18	СН	2-CH ₂ COOMe	NMe ₂
	19	11	2-CH ₂ COOEt	"
	20	.,	2-СH ₂ СООН	. 11
10	. 21	n	2-CH ₂ COOMe	NMe ₂
	22	11	2-сн ₂ соон	n N
	23	ti .	2-CH ₂ COOMe	N
15	24	11	2-сн ₂ соон	"
	25	11	2-CH ₂ COOMe	N NMe
	26	11	2-сн ₂ соон	п
20	27	11	2-CH ₂ CH ₂ COOMe	✓ NMe ₂
	. 28	H	2-СH ₂ СH ₂ СООН	. "
	29	ŧŧ	3-CH ₂ COOMe	u ,
25	30	11	3-СН ₂ СООН	"
	31	. "	2-СH ₂ CH ₂ OH	u .
	32	11	2-CH ₂ CH ₂ OC(Ph) ₃	u .
30	33	11	2-СH ₂ СH ₂ СH ₂ ОН	н
	34	N	2-COOMe	NEt ₂
	35	11	2-COOH	и
35	36	11	2-CH ₂ COOMe	✓ NMe ₂
	37	F 5	2-сн ₂ соон	п
:		•		

	Compound No.	х	-Y-A	-(CH ₂) _n -z
5	38	N	2-CH ₂ COOMe	NEt ₂
	39	11	2-сн ₂ соон	"
	40	n	2-CH ₂ COOMe	✓ NMe ₂
10	41	11	2-сн ₂ соон	
	42	11	2-CH ₂ CH ₂ COOMe	NEt ₂
	43	. 11	2-сн ₂ сн ₂ соон	и .
15	44	11	2-СН (СН ₃) СООМе	✓ NMe ₂
	45	11	2-СН (СН ₃) СООН	11
	46	11	3-CH ₂ COOMe	u
20	47	11	3-сн ₂ соон	"
	48		3-CH ₂ COOMe	NMe ₂
	49	. 11	3-сн ₂ соон	11
25	50	СН2	2-C00Me	NMe ₂
	51	11	2-СООН .	II
	52	11	2-СН ₂ СООН	п
30	53	СН	2 - N	NMe ₂
	54	. СН ₂	2-\(\int_{N}\)	n
			W- I	
35			,	
			. 1	

Table 3

	Compound	Chemical shift	Measure solvent	
10		Cis	Trans	sorvenc
	1	5.67	6.06	A
	2	5.70	6.07	Α ·
15	3	5.72	6.09	В
	4	5.69	6.05	A
	5	5.73	-	В
	6	5.70	6.07	A
20	7	5.71	6.09	В
	8	5.70	6.08	· A
	9	5.71	6.08	В
	10	5.85	6.22	A
25	11	_	6.11	В
	12	5.81	6.20	A
	13	5.81	6.13	В
30	14	5.81	6.18	A
30	15	5.80	6.13	В
	16	5.83	6.19	A
	17	5.92	6.28	A
35	18	5,69	6.06	A
	19	5.70	6.07	Α
	20	5.66	6.00	В

	Compound	Chemical shift	Measure	
_	Compound	Cis	Trans	solvent
5	21	5.66	6.02	A
	22	5.67	6.02	В
	23	5.69	5.99	A
10	24	5.60	5.92	A
	25	5.84	6.17	A
	26	5.72	6.05	В
:	27	5.69	6.57	A ·
15	28	5.50	5.99	В
	31	5.66	5.99	A
	32	5.69	6.97	A
	33	5.65		A
20		<u> </u>		

 $A = CDC1_3$

 $B = DMSO-d_6$

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Table 4

5	Compound	Retention tim	Eluent	
	Compound	Cis	Trans	Erdenc
	3	10.33	8.33	В
	5	7.19	6.06	С
10	7	10.83	8.79	В
	9	14.26	11.40	В
	11	27.06	21.33	A
	13	16.59	13.13	A
15	15	_	14.73	A
	20	9.93	7.46	В
	22	11.10	8.40	В
20	24	10.50	8.00	В
. 20	26	11.20	8.93	В
	28	11.60	9.10	В
	33	11.06	-	В

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Instrument: SHIMAZU LC-3A

Condition :

Column Yamamurakagaku YMC A-312 Eluent Ion pair three kinds

A 0.01M PIC B-8 in 54.3% MeOH

B 0.01M PIC B-8 in 61.3% MeOH

C 0.01M PIC B-8 in 66.0% MeOH

* PIC : PIC reagent (Produced by Water

Associates)

35 Pressure : $85-95 \text{ kg/cm}^2$

Temperature: room temperature

Compound (I) has an antiallergic activity and/or antiinflammatory activity. Among Compound (I), the compound represented by the formula (I') has strong antiallergic activity and the compound represented by the formula (I") has strong antiinflammatory activity.

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In the formula, X, n and Z are as previously defined, -Y'-A'' is -Y-A when X is $= CH - \text{ or } -CH_2 - \text{ and is}$ 15 -Y-A which is bound at 2 position of the mother nucleus when X is = N-, and Y and A are as previously defined.

In the formula, n and Z are as previously

defined; Y" is -CH₂- or -CHR₃- substituted at 2 or 3

position of the mother nucleus wherein R₃ is a lower

alkyl; A'" is a hydroxymethyl, a loweralkoxymethyl, a

triphenylmethyloxymethyl, a lower alkanoyloxymethyl, a

formyl, a carboxyl, a lower alkoxycarbonyl, a triphenyl
methyloxycarbonyl, -CONR₁R₂ wherein R₁ and R₂ are the

same or different and are hydrogen atom or a lower alkyl,

4,4-dimethyl-2-oxazoline-2-yl or -CONHOH.

The antiallergic activity and antiinflammatory activity of Compound (I) are described below:

Test for antiallergic activity:

Antiallergic activity was investigated by a homologous PCA (passive cutaneous anaphlaxis) of rats for 48 hours, where Wistar male rats having body weights of 180 to 220 g were used for sampling of antiserum and Wistar male rats having body weights of 120 to 140 g were used for the PCA test.

A) Preparation of anti EWA rat serum

Anti-egg white albumin (EWA) rat serum was prepared according to Stotland and Share's method [Canad. J. Physiol. Pharmacol. 52, 1114 (1974)]. That is, 1 mg of EWA was mixed with 20 mg of aluminum hydroxide gel and 0.5 ml of mixed vaccine of pertussis, diphtheria and tetanus, and the mixture was subcutaneously administered in four portions into rat's footpad. After 14 days, blood was sampled from the carotid artery, and the serum was separated from the sampled blood, and preserved under freezing at -80°C. The potency of the antiserum in the homologous PCA for 48 hours was 1:32.

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B) Homologous PCA test of rats for 48 hours Groups each consisting of 3 rats were used, and 0.05 ml of anti-EWA rat serum diluted with a physiological saline solution to 8 times as much was intradermally injected each at two positions of dorsal skin to make the animals passively sensitised. After 47 hours, the compound of the present invention, or its solution (physiological saline solution or CMC solution) was orally administered. One hour thereafter, 0.5 ml/100 g of 1% Evan's blue physiological saline solution containing 2 mg of the antigen EWA was administered into the tail vein, and 30 minutes thereafter, the animals were sacrificed by exsanguination. Then, the skins were removed and the amount of leaked dye at the blue-dyed parts was measured according to the Katayama et al method [Microbiol. Immunol. 22, 89 (1978)]. That is, the blue-dyed parts were cut out by scissors, and placed in test tubes containing 1 ml

of 1N KOH and incubated at 37°C for 24 hours. Then, 9 ml of a mixture of 0.6N phosphoric acid and acetone (5:13) was added thereto, and the mixture was shaked and centrifuged at 2,500 rpm for 10 minutes. Absorbancy of the supernatant at 620 µm was measured, and the amount of leaked dye was quantitatively determined by the calibration curve prepared in advance. An average of measurements at the two position was made a value for one zooid, and inhibition rate for the individual zooid was calculated by the following formula:

Inhibition rate (%) =

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Average leaked amount Leaked amount of of solvent-admini- - test compound- stered group administered group x 100

Average leaked amount of solvent-administered group

Cases where, the inhibition rate is 50% or higher, were regarded as positive PCA inhibition activity, and the minimum administered dosage, where a positive case was observed in at least one of three zooids was regarded as minimum effective dosage (MED). The results are shown in Table 5.

Acute toxic test:

Groups each consisting of 3 dd, male mice having body weights of 20 ±1 g were used, and the compound of the present invention was administered orally (po: 300 mg/kg) or intraperitoneally (ip: 100 mg/kg). Mortality 7 days after the administration was observed to obtain MLD (minimum lethal dosage). The results are shown in Table 5.

Antiinflammatory activity test:

Antiinflammatory activity was examined according to Rat carageenin paw edema [J. Pathol. 104, 15-29 (1971)], Groups each consisting of three Wistar male rats weighing 150 g were used. The test compound was suspended in 0.3% aqueous CMC solution and the suspension was given orally.

Sixty minutes later, 0.1 ml of 0.1% carageenin was subcutaneously injected in a hind paw to form carageenin paw edema.

The volume of paw was measured by dipping the paw to the extent of the part right over fibula malleolus lateralis before the administration and 3 hours after the administration of carageenin with plethysmometer.

The ratio of the volume 3 hours after the administration to that before the administration of carageenin was calculated and each ratio is compared with the ratio of control group (0.3% CMC was administered) to give the edema inhibiting percentage. The results are shown in Table 6.

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Table 5

5	Compound	Acute toxicity (MLD) mg/kg		Antiallergic Activity Number of positive zooids in one group of 3 zooids Dosage						M E D
	·	po	ip	100	10	1.	0.1	0.01	0.001	57 5
	3 (cis)	>300	>100	3/3	3/3	3/3	3/3	0/3	 .	0.1
10	3' (trans)	>300	>100	3/3	2/3	1/3	1/3	0/3	-	0.1
	5' (cis)	>300	>100	3/3	3/3	3/3	0/3	0/3	_	1
15	7' (cis:trans = 7:3)	>300	>100	3/3	2/3	1/3	0/3	_	-	1
	9 (cis:trans = 91:9)	>300	>100	3/3	3/3	2/3	0/3	0/3	-	1
20	ll' (trans)	>300	>100	2/3	1/3	0/3	0/3	. –	· -	10
	13' (cis:trans = 7:93)	>300	>100	3/3	1/3	0/3	0/3	_	_	10
25	15' (trans)	-	-	3/3	0/3	0/3	0/3	_	-	100
	20' (trans)	>300	>100	3/3	3/3	3/3	1/3	0/3	·	0.1
30	20 (trans)	>300	>100	2/3	2/3	3/3	3/3	0/3	0/3	0.1
•	20 (cis)	>300	>100	3/3	3/3	3/3	3/3	1/3	0/3	0.01
35	22 (cis:trans = 92:8)	>300	>100	3/3	3/3	2/3	1/3	0/3	_	0.1
	26' (cis:trans = 12:88)	>300	>100	3/3	3/3	2/3	0/3			1

	28' (cis:trans = 37:63)	>300	>100	3/3	3/3	3/3	2/3	2/3	0/3	0.01
5	28 (cis)	>300	>100	3/3	2/3	3/3	1/3	0/3		0.1
	28 (trans)	>300	>100	3/3	3/3	2/3	2/3	1/3	0/3	0.01
10	31' (trans)	>300	>100	3/3	3/3	3/3	1/3	0/3	-	0.1
	31 (trans)	>300	>100	3/3	3/3	2/3	3/3	0/3	-	0.1
15	31 (cis)	200	>100	-	3/3	3/3	2/3	0/3	0/3	0.1
	33' (cis)	NT	NT	3/3	3/3	1/3	0/3	. 		1
20	35' (syn:anti = 1:1)	300>	100>	3/3	1/3	0/3	-	-	<u>-</u>	10
	37 (syn:anti = 8:92)	300>	100>	3/3	3/3	0/3	-	-	-	10
0.5	39 (syn:anti = 2:98)	300>	100>	3/3	2/3	3/3	0/3	-	-	1
25	41 (syn:anti = 3:97)	300>	100>	3/3	2/3	. 1/3	0/3	- ,		1
	43' syn:anti mixture	300>	100>	3/3	2/3	0/3	0/3	· -	-	10
30	45' (anti)	300>	•		3/3	2/3	0/3	<u>.</u>	- <u>-</u>	1
						-		-		• • • • • • • • • • • • • • • • • • • •

Table 6

5	Compound No.	Carageenin paw edema inhibiting percentage (%) (Average value in one group of 3 rats, 100 mg/kg oral administration)
	37	51.6
	39	50.2
10 .	41	38.7
	45'	63.1
	47	46.0
	49	24.1
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As is evidenced in Tables 5 and 6, Compound (I) 20 and pharmaceutically acceptable salt thereof have PCA inhibiting activity and/or carageenin paw edema inhibiting activity.

PCA inhibiting activity is believed to be on the basis of an activity inhibiting liberation of chemical mediator such as histamine from rat skin cell. Therefore, Compound (I) and pharmaceutically acceptable salts thereof are believed to be useful for treating an allergic disease such as bronchus asthma which is caused by trachea contractile activity of chemical mediator such as histamine.

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On the other hand, carageenin paw edema inhibiting activity is believed to be on the basis of prostaglandin biosynthesis inhibiting activity. Thus, Compound (I) and pharmaceutically acceptable salts thereof are believed to be useful for treating an acute inflammation and rheumatism which are ascribed to excessive prostaglandin.

Compound (I) includes a compound having both antiallergic and antiinflammatory activities described above which is useful for the treatment of allergic diseases accompanied by inflammation.

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In view of the pharmacological activity of Compound (I), Compound (I) can be used in various medicament forms for the administration purposes.

The present medicament composition can be prepared by uniformly mixing an effective amount of a free Compound (I) or a pharmaceutically acceptable salt thereof as an active component with a pharmaceutically acceptable The carrier can take a wide range of forms in accordance with a desirable medicament form for the admi-These medicament compositions are desirably in a unit dosage form suitable for the oral administration or injection administration. In the preparation of a composition in the oral dosage form, any useful, pharmaceutically acceptable carrier can be used. For example, an oral liquid preparation such as a suspended medicament or syrup medicament can be prepared using water; such as sucrose, sorbitol, fructose, etc.; glycols such as polyethylene glycol, propylene glycol, etc.; oils such as sesame oil, olive oil, soybean oil, etc.; antiseptics such as alkyl parahydroxybenzoate, etc.; and flavors such as strawberry flavor, peppermint, etc. Powder, pills, capsules and tablets can be prepared using an excipient such as lactose, glucose, sucrose, mannitol, etc.; a disintegrator such as starch, sodium alginate, etc.; lubricant such as magnesium stearate, talc, etc.; such as polyvinyl alcohol, hydroxypropylcellulose, gelatin, a surfactant such as fatty acid esters; and a plasticizer such as glycerine, etc. Tablets and capsules are the most useful, oral unit dosage forms because of easy administration. To prepare tablets and capsules, solid carriers for medicament are used. Injection solution can be prepared using a carrier consisting of a salt solution, a glucose solution or a mixture of the salt solution and the glucose solution. The effective dosage of Compound (I) is 1 to 20 mg/kg/day for a human being, and number of administration is 3 to 4 per day.

Examples and Reference Examples are given below:

Reference example 1

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(Raw material 1) Methyl ll-oxo-6,ll-dihydrodibenz[b,e] oxepin-2-carboxylate

In this example, 348.9 g of sodium salt of methyl p-hydroxybenzoate, 402.4 g of phthalide and 200 g of sodium chloride are mixed with one another and stirred at 150°C for 6 hours. After completion of the reaction, the mixture is cooled until the temperature is brought back to room temperature, 4 % of aqueous 10 % acetic acid solution is added thereto and the mixture is allowed to stand at room temperature overnight. After stirring the mixture at room temperature for 3 hours, deposited crystals are separated by filtration, and 6 & of water is added thereto. After stirring the mixture at room temperature for 30 minutes, the deposited crystals are separated by filtration. After the addition of 3.1 of toluene to the crystals, the mixture is stirred at room temperature for one hour. The crystals are separated by filtration and dried over heating under reduced pressure to yield 393.9 g of 2-(4-methoxycarbonylphenoxy) benzoic acid methyl ester.

TR (KBr disk): 3400, 1700, 1610, 1260, 1235 cm⁻¹
The thus obtained 2-(4-methoxycarbonylphenoxy) benzoic acid methyl ester (392.7 g) is suspended in 5.0 l of methylene chloride and 266.0 g of trifluoroacetic anhydride is added thereto. After stirring the mixture at room temperature for one hour, 19.4 g of boron trifluoride-ethylether complex is added thereto and the mixture is stirred at room temperature for two hours. The reaction solution is poured into ice water. After an organic solvent layer is separated from the mixture, the organic layer is washed with diluted aqueous sodium hydroxide solution and water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 335.3 g of methyl 11-oxodibenz[b,e]oxepin-2-carboxylate as a white crystal.

Melting point and elementary analysis are shown in Table 7.

IR (KBr disk): 1710, 1650, 1610, 1250, 1010 cm $^{-1}$

NMR (CDCl₃, δ , ppm): 3.84(s, 3H), 5.14(s, 2H), 6.87-8.93(m, 7H)

Reference examples 2 - 5

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- 5 (Raw material 2) 11-0xo-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid
 - (Raw material 3) 11-0xo-6,11-dihydrodibenz[b,e]oxepin-3-acetic acid
 - (Raw material 4) 2-(11-0xo-6,11-dihydrodibenz[b,e]oxepin-2-y1)-propionic acid
 - (Raw material 5) 3-(ll-Oxo-6,ll-dihydrodibenz[b,e]oxepin-2-yl)-propionic acid

Raw materials 2-5 are produced by respectively substituting p-hydroxyphenyl acetic acid, m-hydroxyphenyl acetic acid, 2-(p-hydroxyphenyl)-propionic acid and 3-(p-hydroxyphenyl)-propionic acid for methyl p-hydroxybenzoate in Reference example 1.

Melting points and elementary analyses thereof are shown in Table 7.

Reference example 6

(Raw material 6) Methyl 11-methylene-6,11-dihydrodibenz-[b,e]oxepin-2-carboxylate

of methyltriphenylphosphonium bromide and 40 ml of 1.6 Nn-butyl lithium hexane solution is dropwise added thereto
under a nitrogen atmosphere and ice-cooling.

After stirring the mixture under ice-cooling for 30
minutes, a solution of 15 g of methyl ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate in 250 ml of tetrahydrofuran is dropwise added thereto and the mixture is stirred
at room temperature for two hours. The solvent is distilled away under reduced pressure and the residue is
purified by column chromatography on silica gel (eluent:
hexane: ethyl acetate = 3:1) to obtain 3.7 g of the

desired product as a colorless oily matter.

NMR (CDCl $_3$, δ , ppm): 3.83(s, 3H), 5.15(s, 2H), 5.29 (s, 1H), 5.74(s, 1H), 6.69-8.22(m, 7H) Melting point and elementary analysis are shown in Table 7.

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Reference example 7

(Raw material 7) Methyl ll-methylene-6,ll-dihydrodibenz-[b,e]oxepin-2-acetate

The desired product is obtained by substituting 10 ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid for methyl ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate in Reference example 6.

Colorless oily matter

NMR (CDCl₃, δ, ppm): 3.48(s, 2H), 3.61(s, 3H), 5.05(s, 2H), 5.20(s, 1H), 5.62(s, 1H), 6.59-7.43(m, 7H)

IR (neat, cm⁻¹): 2950, 1740, 1615, 1490, 1010

Melting point and elementary analysis are shown in Table 7.

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Reference example 8

(Raw material 8) 11-Methylene-6,11-dihydrodibenz[b,e]-oxepin-2-acetic acid

To a mixed solvent of 200 ml of methanol and 50 ml of 2N-aqueous sodium hydroxide solution is added 2.9 g of methyl 11-methylene-6,11-dihydrodibenz[b,e]-oxepin-2-acetate (raw material 7, Reference example 7) and the mixture is heated at reflux for two hours. After allowing the mixture to stand for cooling, the mixture is concentrated under reduced pressure, and the pH of the mixture is adjusted to 1.0 with aqueous 4N-hydrochloric acid. The mixture is extracted with 500 ml of ethyl acetate, washed with aqueous lN-hydrochloric acid solution and saturated aqueous sodium chloride solution in order and dried over anhydrous sodium sulfate. The solvent is distilled away under reduced pressure and the resultant crude product is crystallized from hexane to obtain 2.7 g

of the desired product as a white solid.

NMR (DMSO-d₆ + D₂O, δ , ppm): 3.45(s, 2H), 5.02(s, 2H), 5.16(s, 1H), 5.60(s, 1H), 6.45-7.44(m, 7H) Melting point and elementary analysis are shown in Table 7.

Reference example 9

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(Raw material 9) Methyl ll-methylene-6,ll-dihydrodibenz-[b,e]oxepin-3-acetate

The desired product is obtained by substituting ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-3-acetic acid for methyl ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate in Reference example 6.

15 Reference example 10

(Raw material 10) 11-Methylene-6,11-dihydrodibenz[b,e] oxepin-3-acetic acid

The desired product is obtained by substituting methyl ll-methylene-6,ll-dihydrodibenz[b,e]oxepin-3-acetate for methyl ll-methylene-6,ll-dihydrodibenz[b,e]oxepin-2-acetate in Reference example 8.

Table 7

25	Raw material	Melting point (°C)	Elementary a	s (%)	
	1	128 - 129	as C ₁₆ H ₁₂ O ₄		
			•	С	Н
30		(Isopropyl	Calculated	71.63	4.51
		ether)	Found	71.55	4.48
	2 .	130 - 132	as C ₁₆ H ₁₂ O ₄		
				С	Н
35	·	(Ethyl	Calculated	71.63	4.51
		acetate)	Found	71.86	4.55
		<u></u>			

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	Raw material	Melting point (°C)	Elementary analysis (%) or mass spectrum			
	3	111 - 114	as C ₁₆ H ₁₂ O ₄			
5.			С н			
		(Ethyl	Calculated 71.63 4.51			
•		acetate)	Found 71.53 4.66			
	4	Syrup	as C ₁₇ H ₁₄ O ₄			
10			(M + 282)			
	5	144 - 145	as C ₁₇ H ₁₄ O ₄			
			с н			
· •	, , , l	(Water)	Calculated 72.33 5.00			
15			Found 72.45 5.20			
13	6	Syrup	as C ₁₇ H ₁₄ O ₃			
	•		(M + 266)			
	. 7	Syrup	as C ₁₈ H ₁₆ O ₃			
20			(M + 280)			
	8	162 - 163	as C ₁₇ H ₁₄ O ₃			
i	•	·	С н			
		(Water)	Calculated 76.68 5.30			
25	, ,		Found 76.29 5.16			
د ے		1				

Reference example 11

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(Reagent 1) (3-Dimethylaminopropyl)-triphenylphosphonium bromide hydrobromide

In this example, 350.0 g of triphenylphosphine and 270.0 g of dibromopropane are suspended in 700 ml of toluene and the suspension is heated at reflux for 25 hours. After allowing the suspension to stand for cooling, the formed product is separated by filtration and washed with 2 l of toluene to obtain 550.0 g of (3-bromopropyl)-triphenylphosphonium bromide hydrobromide having m.p. 233 - 234°C.

Then, 100.0 g of (3-bromopropyl)-triphenyl-phosphonium bromide hydrobromide is suspended in 500 ml of ethanol and 300 ml of 50 % aqueous dimethylamine solution is added thereto. After heating the mixture at reflux for 10 minutes, the mixture is allowed to stand for cooling. The solvent is distilled away under reduced pressure and the resultant crude product is recrystallized from ethanol to obtain 64.0 g of the desired product having the physicochemical properties as identified in Table 8.

Reference examples 12 - 14

- (Reagent 2) (3-Diethylaminopropyl)triphenylphosphonium bromide hydrobromide · 1/3 hydrate
- 15 (Reagent 3) (4-Dimethylaminobutyl)triphenylphosphonium bromide hydrobromide
 - (Reagent 4) (3-Pyrrolidinopropyl)triphenylphosphonium bromide hydrobromide · 1/2 hydrate

The above-captioned compounds are prepared

20 according to the same manner as described in Reference
example 11 and the physicochemical properties are shown in
Table 8.

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Table 8

į	Reagent	Melting point (°C)	Elementary	analysi	s (%)		
5	1	287 – 289	as C ₂₃ H ₂₈ NP	Br ₂			
		(Ethanol)		· C	Н	N	
			Calculated	54.24	5.54	2.75	
			Found	54.12	5.63	2.93	
10	2	228 - 230	as C ₂₅ H ₃₂ NPBr ₂ · 1/3H ₂ O				
		(Isopropanol)	-	C	H	N	
		(200p20pano2)	Calculated	55.33	6.05	2.58	
	٠.		Found	55.31	6.19	2.68	
15	3	255 - 257	as C ₂₄ H ₃₀ NP	Br ₂	-		
		(Isopropanol)	•	C	Н	N	
			Calculated	55.09	5.78	2.68	
-	•. •		Found	55.04	5.91	2.62	
•	4	291 - 293	as C ₂₅ H ₃₀ NP	Br ₂ · 1/	2H ₂ O		
20		(Ethanol)		С	Н	N	
			Calculated	55.17	5.74	2.57	
			Found	55.18	5.95	2.66	
			•				

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Example 1

Ethyl 11-(3-dimethylaminopropylidene)-6,11-dihydro-dibenz[b,e]oxepin-2-carboxylate (Compound 2)

Process A:

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N-(1,1-dimethy1-2-hydroxyethy1)-11-oxo-6,11-dihydro-dibenz[b,e]oxepin-2-carboxamide

In this process, 12.5 g of 6,ll-dihydro-ll-oxodibenz[b,e]oxepin-2-carboxylic acid is dissolved in 300 ml of methylene chloride and 8.9 g of thionyl chloride is dropwise added to the solution under ice-cooling. After stirring the mixture at room temperature for two hours, the solvent is distilled away under reduced pressure. To the obtained residue are added 100 ml of toluene and 32.4 g of 2-amino-2-methyl-propanol, and the mixture is stirred at 50°C for 3 hours.

The mixture is extracted with 500 ml of ethyl acetate, and washed with saturated aqueous sodium bicarbonate solution and saturated aqueous sodium chloride solution in order. The mixture is dried over anhydrous sodium sulfate and the solvent is distilled away under reduced pressure. The crude product is recrystallized from toluene to obtain 8.3 g of the desired product as a white crystal.

Melting point: $155 - 159^{\circ}$ C

NMR (CDCl₃ + DMSO-d₆, δ , ppm): 1.38(s, 6H), 3.53(s, 2H), 5.25(s, 2H), 6.91-8.68(m, 7H)

Process B:

2-(4,4-Dimethyl-2-oxazoline-2-yl)-ll-oxo-6,ll-dihydrodibenz[b,e]oxepin

In this process, 8.0 g of N-(1,1-dimethy1-2-hydroxyethy1)-ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-carboxamide is suspended in 100 ml of methylene chloride. To the suspension is added 3.6 g of thionyl chloride under a nitrogen atmosphere and ice-cooling and the mixture is stirred at room temperature for one hour. To the mixture

is added 300 ml of methylene chloride, and the mixture is washed with saturated aqueous sodium bicarbonate solution and dried over anhydrous magnesium sulfate. The solvent is distilled away under reduced pressure and the residue is purified by column chromatography on silica gel (eluent: hexane: ethyl acetate = 2:1). The resultant crude product is recrystallized from hexane to obtain 6.3 g of the desired product as a white crystal.

Melting point: 122°C

NMR (CDCl₃, δ , ppm): 1.37(s, 6H), 4.06(s, 2H), 5.14(s, 2H), 6.84-8.89(m, 7H)

Elementary analysis (%): as C₁₉H₁₇O₃N

Calculated: C 74.25 H 5.58 N 4.56

Found: C 74.23 H 5.55 N 4.59

Process C:

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11-(3-Dimethylaminopropyl)-11-hydroxy-2-(4,4-dimethyl-2-oxazoline-2-yl)-6,11-dihydrodibenz[b,e]oxepin

To a solution of 3-dimethylaminopropyl magnesium chloride obtained by reacting 1.2 g of magnesium with 6.0 g of 3-dimethylaminopropyl chloride in 80 ml of tetrahydrofuran under a nitrogen atmosphere using dibromoethane as a catalyst is dropwise added under ice-cooling 80 ml of tetrahydrofuran solution of 7.6 g of 2-(4,4-dimethyl-2-oxazoline-2-yl)-ll-oxo-6,ll-dihydrodibenz[b,e]oxepin.

After stirring the mixture at room temperature overnight, aqueous ammonium chloride solution is added thereto and then the mixture is neutralized with aqueous 4N-hydrochloric acid solution. The solvent is distilled away under reduced pressure. To the residue is added aqueous 4N-hydrochloric acid solution to adjust the pH of the solution to 1. After washing the mixture with 200 ml of diethyl ether, aqueous 10N-sodium hydroxide solution is added to adjust the pH of the mixture to 13. The mixture is extracted with 200 ml of methylene chloride and the extract is washed with saturated aqueous sodium

bicarbonate solution and saturated aqueous sodium chloride

solution in order. After drying the solution over anhydrous sodium sulfate, the solvent is distilled away under reduced pressure. The residue is purified by column chromatography on silica gel (eluent: hexane: ethyl acetate: triethylamine = 10:10:1). The resultant crude product is triturated with isopropyl ether to obtain 6.1 g of the desired product as a white solid.

Melting point: 166 - 167°C

NMR (CDCl₃, δ, ppm): 1.30(s, 8H), 2.18(s, 8H), 3.98 (s, 2H), 4.97 and 5.46(ABq, J=15.1 Hz, 2H), 6.65-8.49(m, 7H)

Process D:

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Ethyl 11-(3-dimethylaminopropylidene)-6,11-dihydro-dibenz[b,e]oxepin-2-carboxylate

In this process, 6.1 g of 11-(3-dimethylamino-propyl)-11-hydroxy-2-(4,4-dimethyl-2-oxazolin-2-yl)-6,11-dihydrodibenz[b,e]oxepin is dissolved in 300 ml of ethanol. To the solution are added 0.6 g of p-toluenesulfonic acid and 30 ml of water and the mixture is heated at reflux for 4 hours. The solvent is distilled away under reduced pressure to obtain a crude product of 11-(3-dimethylamino-propylidene)-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid. The crude product is dissolved in 300 ml of ethanol and 20 ml of concentrated sulfuric acid is added thereto. The mixture is heated at reflux for 15 hours.

The solvent is distilled away under reduced pressure. To the resultant residue is added 200 ml of water and the mixture is washed with diethyl ether. The pH of the mixture is adjusted to 12.0 with aqueous 10N-sodium hydroxide solution and the mixture is extracted with 300 ml of methylene chloride. The extract is washed with saturated aqueous sodium bicarbonate solution and saturated aqueous sodium chloride solution in order. After drying the extract over anhydrous sodium sulfate, the solvent is distilled away under reduced pressure and

the solvent is distilled away under reduced pressure and the resultant residue is purified by column chromatography

on silica gel (eluent: ethyl acetate: triethylamine = 10:1) to obtain 1.4 g of the desired product as a color-less oily matter.

IR (neat, cm^{-1}): 2950, 2775, 1715, 1250, 1120, 1010 Mass spectrum (m/z): 351 (M⁺)

Example 2

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11-(3-Dimethylaminopropylidene)-2-(2-triphenyl-methyloxymethyl)-6,11-dihydrodibenz[b,e]oxepin (Compound 32)

Process A:

11-Hydroxy-2-(2-hydroxyethyl)-6,11-dihydrodibenz
[b,e]oxepin

In this process, 20 g of methyl ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-acetate is dissolved in 500 ml of tetrahydrofuran. To the solution is added 6.0 g of lithium alminium hydroxide and the mixture is stirred at room temperature for one hour. After decomposing an excess of the reagent by the addition of water to the mixture, the mixture is filtered to remove an inorganic salts and the filtrate is concentrated to dryness under reduced pressure to obtain 17.7 g of the desired product as a white solid.

Melting point: 132 - 136°C

NMR (CDCl₃ + DMSO-d₆ + D₂O, δ , ppm): 2.59(t, 2H, J= 6.8Hz), 3.55(t, 2H, J=6.8Hz), 4.89 and 5.71(ABq, 2H, J=12.6Hz), 5.60(s, 1H), 6.46-7.49(m, 7H)

Process B:

11-Hydroxy-2-(2-triphenylmethyloxyethyl)-6,11-dihydrodibenz[b,e]oxepin

In this process, 17.2 g of ll-hydroxy-2-(2-hydroxyethyl)-6,ll-dihydrodibenz[b,e]oxepin is dissolved in 50 ml of pyridine. To the solution is added 30 g of triphenylchloromethane and the mixture is stirred at 50°C for 5 hours. After adding water and stirring the mixture

for 2 hours, the solvent is distilled away under reduced pressure. The mixture is extracted with 1000 ml of ethyl acetate, washed with saturated aqueous sodium chlroide solution, and dried over anhydrous sodium sulfate. The solvent is distilled away under reduced pressure and the resultant residue is purified by column chromatography on silica gel (eluent: hexane: ethyl acetate = 3:1) to obtain 21.7 g of the desired product as a colorless amorphous powder.

NMR (CDCl₃ + D₂O, δ , ppm): 2.47-2.95(m, 2H), 2.96-3.45(m, 2H), 4.87 and 5.71(ABq, 2H, J=13.2Hz), 5.43(s, 1H), 6.33-7.51(m, 22H)

Process C:

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11-0xo-2-(2-triphenylmethyloxyethyl)-6,ll-dihydrodibenz[b,e]oxepin

In this process, 10 g of 11-hydroxy-2-(2-triphenylmethyloxyethyl)-6,ll-dihydrodibenz[b,e]oxepin is dissolved in a solution comprising 800 ml of acetone, 1000 ml of water, 20 ml of saturated aqueous magnesium sulfate solution and 0.2 g of disodium phosphate. solution is dropwise added 2.6 g of aqueous sodium permanganate solution and the mixture is stirred at room temperature for 4.5 hours. Then, 100 ml of methanol is added thereto and the mixture is heated at reflux for 3 After allowing the mixture to stand for cooling, the mixture is filtered and the filtrate is extracted with 1000 ml of ethyl acetate, washed with saturated aqueous sodium chloride solution and dried over anhydrous sodium sulfate. The solvent is distilled away under reduced pressure and the resultant crude product is recrystallized from isopropanol to obtain 8.0 g of the desired product having melting point of 132 -134°C as a white crystal.

Elementary analysis (%): as C₃₅H₂₈O₃
Calculated: C 84.65 H 5.68
Found: C 84.56 H 5.67

NMR (CDCl₃, δ, ppm): 2.61-3.04(m, 2H), 3.05-3.46
(m, 2H), 5.01(s, 2H), 6.63-8.07(m, 22H)

Process D:

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11-(3-Dimethylaminopropyl)-11-hydroxy-2-(2-triphenylmethyloxyethyl)-6,11-dihydrodibenz[b,e]oxepin

To a solution of 3-dimethylaminopropyl magnesium chloride obtained by reacting 0.2 g of magnesium with 1.0 g of 3-dimethylaminopropyl chloride in 10 ml of tetrahydrofuran under a nitrogen atmosphere using dibromoethane as a catalyst, is dropwise added a solution obtained by dissolving 2.0 g of 11-oxo-2-(2-triphenylmethyloxyethyl)-6,ll-dihydrodibenz[b,e]oxepin in 10 ml of tetrahydrofuran under ice cooling and the mixture is stirred at room temperature for one day. Aqueous ammonium chloride solution is added thereto and the pH of the mixture is adjusted to 7.0 with aqueous 4N-hydrochloric acid solution. solvent is distilled away under reduced pressure. mixture is extracted with 200 ml of methylene chloride and washed with saturated aqueous sodium bicarbonate solution and saturated aqueous sodium chloride solution in order. After drying the extract over anhydrous sodium sulfate, the solvent is distilled away under reduced The resultant residue is purified by column chromatography on silica gel (eluent: hexane : ethyl acetate : triethylamine = 10:10:1) to obtain 1.2 g of

NMR (CDCl₃, δ, ppm): 0.85-1.83(m, 4H), 2.08(s, 6H), 2.67-3.44(m, 6H), 4.94 and 5.36(ABq, 2H, J= 15.8Hz), 6.63-8.13(m, 22H)

Mass spectrum (m/z): 583 (M⁺)

desired product as a colorless amorphous powder.

Process E:

11-(3-Dimethylaminopropylidene)-2-(2-triphenylmethyloxyethyl)-6,11-dihydrodibenz[b,e]oxepin

In this process, 1.2 g of 11-(3-dimethylamino-propyl)-11-hydroxy-2-(2-triphenylmethyloxyethyl)-6,11-dihydrodibenz[b,e]oxepin is dissolved in 50 ml of pyridine. To the solution is dropwise added 0.8 g of phosphorus oxychloride under a nitrogen atmosphere and ice-cooling.

After stirring the mixture at room temperature for one hour, the solvent is distilled away under reduced pressure. The residue is extracted with 100 ml of methylene chloride, and washed with saturated aqueous sodium bicarbonate solution and saturated aqueous sodium chloride solution in order. After drying the mixture over anhydrous sodium sulfate, the solvent is distilled away under reduced pressure. The resultant residue is purified by column chromatography on silica gel (eluent: hexane: ethylacetate: triethylamine = 10:10:1) to obtain 0.82 g of the desired product as a colorless oily matter.

NMR (CDCl₃, δ , ppm): 2.16(s, 6H), 2.30-2.40(m, 4H), 2.79(t, 2H, J=6Hz), 3.24(t, 2H, J=6Hz), 5.97 (t, 1H, J=7Hz), 6.60-7.40(m, 22H), (trans form) Mass spectrum (m/z): 565 (M⁺)

Example 3

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11-(3-Dimethylaminopropylidene)-2-(2-hydroxyethyl)6,ll-dihydrodibenz[b,e]oxepin (Compound 31)

20 In this example, 0.92 g of 11-(3-dimethylaminopropylidene)-2-(2-triphenylmethyloxyethyl)-6,11-dihydrodibenz[b,e]oxepin is dissolved in a mixed solvent of 20 ml of water and 20 ml of dioxane. To the solution is added 60 mg of p-toluene sulfonic acid and the mixture is 25 heated at reflux for two hours. The solvent is distilled away under reduced pressure and the residue is extracted with 200 ml of ethylacetate, washed with saturated aqueous sodium bicarbonate solution and saturated aqueous sodium hydrochloride solution in oder and dried over anhydrous 30 sodium sulfate. The solvent is distilled away under reduced pressure. The resultant residue is purified by column chromatography on silica gel (eluent: ethylacetate : triethylamine = 10 :1) to obtain 0.4 g of the desired product.

2.76(t, 2H, J=6Hz), 3.78(t, 2H, J=6Hz), 5.66(t, 1H, J=7Hz), 6.80-7.40(m, 7H)

Mass spectrum: 323 (M⁺)

Trans form white solid,

Melting point: 96 - 97°C (diethylether)

NMR (CDCl₃, δ, ppm): 2.21(s, 6H), 2.30-2.70(m, 4H), 2.76(t, 2H, J=6Hz), 3.78(t, 2H, J=6Hz), 6.01(t, 1H, J=7Hz), 6.68-7.40(m, 7H)

Mass spectrum (m/z): 323 (M^+)

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Example 4

11-(3-Dimethylaminopropylidene)-6,11-dihydrodibenz
[b,e]oxepin-2-acetic acid (Compound 20)

In this Example, 2.2 g of 11-(3-dimethylamino-propylidene)-2-(2-hydroxyethyl)-6,11-dihydrodibenz[b,e]-oxpein is dissolved in 100 ml of acetone. The Jones reagent is added to the solution until the reaction solution shows an orange color and the mixture is stirred at room temperature for one hour. Sodium bicarbonate is added thereto and an inorganic substance is removed by filtration. The solvent of the filtrate is distilled away under reduced pressure to obtain the desired product. The physicochemical properties of the product are identical with those of the product obtained in Example 35.

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Example 5

Methyl 11-(3-dimethylaminopropylidene)-6,11-dihydro-dibenz[b,e]oxepin-2-carboxylate (Compound 1)

In this Example, 45 g of (3-dimethylaminopropyl)-triphenylphosphonium bromide hydrobromide is suspended in 200 ml of tetrahydrofuran under a nitrogen atmosphere and 82 ml of 1.6N-n-butyl lithium hexane solution is added thereto under ice-cooling. The mixture is stirred under ice-cooling for one hour. To the mixture is dropwise added under ice-cooling a solution obtained by dissolving 10 g of methyl 11-oxo-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate in 200 ml of tetrahydrofuran. After stirring

the mixture at room temperature for 2 hours, the mixture is extracted with 800 ml of ethyl acetate. After washing the extract with saturated aqueous sodium chloride solution and drying the extract over anhydrous sodium sulfate, the solvent is distilled away under reduced pressure. The residue is purified by column chromatography on silicatel (eluent: hexane:ethyl acetate:triethylamine = 10:10:1) to obtain 2.0 g of trans form and 5.6 g of cis form of the desired product.

10 <u>Cis form</u> NMR (CDCl₃, δ, ppm): 2.23(s, 6H),
2.17-2.81(m, 4H), 5.28(bs, 2H), 5.61(t, 1H),
6.80-8.10(m, 7H)

Trans form NMR (CDCl₁, δ, ppm): 2.15(s, 6H).

Trans form NMR (CDCl₃, δ, ppm): 2.15(s, 6H), 2.17-2.81(m, 4H), 5.00-5.50(broad, 2H), 6.06 (t, 1H), 6.70-8.10(m, 7H)

Example 6

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Methyl 11-(3-diethylaminopropylidene)-6,11-dihydro-dibenz[b,e]oxepin-2-carboxylate (Compound 4)

The desired product is obtained by substituting (3-diethylaminopropyl) triphenylphosphonium bromide hydrobromide · 1/3 hydrate for (3-dimethylaminopropyl) - triphenylphosphonium bromide hydrobromide in Example 5.

25 Example 7

Methyl 11-(3-pyrrolidinopropylidene)-6,11-dihydro-dibenz[b,e]oxepin-2-carboxylate (Compound 6)

The desired product is obtained by substituting (3-pyrrolidinopropyl) triphenylphosphonium bromide hydrobromide · 1/2 hydrate for (3-dimethylaminopropyl) triphenylphosphonium bromide hydrobromide in Example 5.

Example 8

Methyl ll-(4-dimethylaminobutylidene)-6,ll-dihydro-dibenz[b,e]oxepin-2-carboxylate (Compound 8)

The desired product is obtained by substituting (4-dimethylaminobutyl) triphenylphosphonium bromide

hydrobromide for (3-dimethylaminopropyl)triphenylphosphonium bromide hydrobromide in Example 5.

Example 9

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Methyl 11-(3-dimethylaminopropylidene)-6,11-dihydro-dibenz[b,e]oxepin-2-acetate (Compound 18)

In this example, 48 g of (3-dimethylaminopropyl)triphenylphosphonium bromide hydrobromide is suspended in 200 ml of tetrahydrofuran under a nitrogen atmosphere and 80 ml of 1.6N-n-butyl lithium hexane solution is added thereto under ice-cooling. The mixture is stirred under ice-cooling for one hour. A solution obtained by dissolving 5.0 g of 11-oxo-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid in 120 ml of tetrahydrofuran is dropwise added under ice-cooling. After stirring the mixture at room temperature for two hours, the solvent is distilled away under reduced pressure. Then, 200 ml of water is added to the residue and the mixture is washed with 200 ml of diethyl ether. The pH of the mixture is adjusted to 1 with aqueous 4N-hydrochloric acid solution and the mixture is washed with diethyl ether.

Then, aqueous 10N-sodium hydroxide solution is added thereto to adjust the pH of the mixture to 7 and the solvent is distilled away under reduced pressure. The resultant residue is dissolved in 400 ml of methanol and 5 g of p-toluene sulfonic acid is added thereto. After heating the mixture at reflux for two hours, the solvent is distilled away under reduced pressure. The residue is extracted with 300 ml of ethyl acetate, washed with saturated aqueous sodium bicarbonate solution and saturated aqueous sodium chloride solution in order and dried over anhydrous sodium sulfate.

The solvent is distilled away under reduced pressure and the resultant residue is purified by column chromatography on silica gel (eluent: hexane:ethyl acetate:triethylamine = 10:10:1) to obtain 4.0 g of the desired product as a colorless oily matter.

Cis form

NMR (CDCl₃, δ , ppm): 2.06-2.67(m, 4H), 2.16(s, 6H), 3.46(s, 2H), 3.58(s, 3H), 5.08(bs, 2H), 5.69 (t, 1H, J=7Hz), 6.53-7.30(m, 7H)

Trans form

NMR (CDCl₃, δ , ppm): 2.06-2.67(m, 4H), 2.16(s, 6H), 3.46(s, 2H), 3.58(s, 3H), 5,08(bs, 2H), 6.06 (t, 1H, J=7Hz), 6.53-7.30(m, 7H)

10 Example 10

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Methyl 11-(4-dimethylaminobutylidene)-6,11-dihydro-dibenz[b,e]oxepin-2-acetate (Compound 21)

The desired product is obtained by substituting (4-dimethylaminobutyl) triphenylphosphonium bromide

hydrobromide for (3-dimethylaminopropyl) triphenylphosphonium bromide hydrobromide in Example 9.

Example 11

Methyl 11-(3-pyrrolidinopropylidene)-6,11-dihydro-dibenz[b,e]oxepin-2-acetate (Compound 23)

The desired product is obtained by substituting (3-pyrrolidinopropyl) triphenylphosphonium bromide hydrobromide ·1/2 hydrate for (3-dimethylaminopropyl) triphenylphosphonium bromide hydrobromide in Example 9.

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Example 12

Methyl 3-[ll-(3-dimethylaminopropylidene)-6,11-dihydrodibenz[b,e]oxepin-2-yl]propionate (Compound 27)

The desired product is obtained by substituting 3-(ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-yl)propionic acid for ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid in Example 9.

Example 13

Methyl 11-(3-dimethylaminopropylidene)-6,11-dihydrodibenz[b,e]oxepin-3-acetate (Compound 29) The desired product is obtained by substituting ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-3-acetic acid for ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid in Example 9.

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Example 14

Methyl 11-(2-dimethylaminoethyl)imino-6,11-dihydrodibenz[b,e]oxepin-2-acetate (Compound 36)

In this example, 22.0 g of methyl 11-oxo-6,11dihydrodibenz[b,e]oxepin-2-acetate and 68.7 g of N,Ndimethylethylenediamine are dissolved in 700 ml of dried To the solution is dropwise added a solution of 17.2 ml of titanium tetrachloride in 40 ml of dried benzene and the mixture is stirred at room temperature overnight. A saturated aqueous sodium bicarbonate solution is added After removing an insoluble solid by filtration, the filtrate is extracted with 500 ml of ethylacetate, washed with saturated aqueous sodium bicarbonate solution and saturated aqueous sodium chloride solution in order, and dried over anhydrous sodium sulfate. The solvent is distilled away under reduced pressure and the residue is purified by column chromatography on silica gel with ethylacetate / triethylamine (10 / 1) as an eluent to obtain 13.8 g of the desired product as a colorless oily matter.

NMR (CDCl₃, δ , ppm): 2.14(s, 6H), 2.63(t, 2H, J= 6.9Hz), 3.51(s, 2H), 3.58(s, 3H), 3.38-3.80 (m, 2H), 5.04(bs, 2H), 6.56-7.60(m, 7H)

IR (neat, cm⁻¹): 2950, 1740, 1630, 1305, 1015

Mass spectrum (m/z): 352 (M⁺)

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Example 15

Methyl 11-(2-diethylaminoethyl)imino-6,11-dihydro-dibenz[b,e]oxepin-2-carboxylate (Compound 34)

The desired product is obtained by substituting methyl ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate for methyl ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-acetate in Example 14 as a colorless oily matter.

Mass spectrum (m/z): 366 (M^+) for $C_{22}H_{26}O_3N_2$

Example 16

Ethyl 11-(2-diethylaminoethyl)imino-6,ll-dihydro-dibenz[b,e]oxepin-2-acetate (Compound 38)

The desired product is obtained by substituting N,N-diethylenediamine for N,N-dimethylenediamine in Example 14 as a colorless oily matter.

Mass spectrum (m/z): 380 (M^+) for $C_{23}H_{28}O_3N_2$

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Example 17

Methyl 11-(3-dimethylaminopropyl)imino-6,11-dihydro-dibenz[b,e]oxepin-2-acetate (Compound 40)

The desired product is obtained by substituting N,N-dimethylpropylenediamine for N,N-dimethylethylenediamine in Example 14 as a colorless oily matter.

Mass spectrum (m/z): 366 (M^{+}) for $C_{22}H_{26}O_{3}N_{2}$

Example 18

Methyl 3-[11-(2-dimethylaminoethyl)imino-6,11dihydrodibenz[b,e]oxepin-2-yl]propionate (Compound 42)

The desired product is obtained by substituting
3-(11-oxo-6,11-dihydrodibenz[b,e]oxepin-2-yl)propionate
acid for methyl 11-oxo-6,11-dihydrodibenz[b,e]oxepin-2acetate in Example 16 as a colorless oily matter.

Mass spectrum (m/z): 394 (M^{+}) for $C_{24}H_{30}O_{3}N_{2}$

Example 19

Methyl 2-[11-(2-dimethylaminoethyl)imino-6,11
dihydrodibenz[b,e]oxepin-2-yl]propionate (Compound 44)

The desired product is obtained by substituting 2-(11-oxo-6,11-dihydrodibenz[b,e]oxepin-2-yl)propionate acid for methyl 11-oxo-6,11-dihydrodibenz[b,e]oxepin-2-acetate in Example 14 as a colorless oily matter.

35 Mass spectrum (m/z): 366 (M^{+}) for $C_{22}H_{26}O_{3}N_{2}$

Example 20

Methyl 11-(2-dimethylaminoethyl)imino-6,ll-dihydro-dibenz[b,e]oxepin-3-acetate (Compound 46)

The desired product is obtained by substituting ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-3-acetic acid for methyl ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-acetate in Example 14 as a colorless oily matter.

Mass spectrum (m/z): 352 (M^+) for $C_{21}H_{24}O_3N_2$

10 Example 21

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Methyl 11-(3-dimethylaminopropyl)imino-6,11-dihydro-dibenz[b,e]oxepin-3-acetate (Compound 48)

The desired product is obtained by substituting ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-3-acetic acid for ll-oxo-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid in Example 17 as a colorless oily matter.

Mass spectrum (m/z): 366 (M^+) for $C_{22}H_{26}O_3N_2$

Example 22

Methyl 11-[2-(4-methylpiperazino)ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 10)

In this example, 1.5 ml of 4-methylpiperazine and 0.37 g of p-formaldehyde are dissolved in 100 ml of tetrachloroethane. To the solution is dropwise added 5 ml of trifluoroacetic acid. After stirring the mixture at 60°C for 2 hours, a solution obtained by dissolving 1.8 g of methyl 11-methylene-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate in 30 ml of tetrachloroethane is dropwise added thereto and the mixture is stirred at 90°C for 3 hours.

The mixture is concentrated to dryness under reduced pressure and aqueous 4N-hydrochloric acid solution is added to the residue to adjust the pH to 1. After washing the solution with diethylether, aqueous 10N-sodium hydroxide solution is added thereto to adjust the pH to 13. The mixture is extracted with 200 ml of methylene chloride, washed with saturated aqueous sodium chloride

solution and dried over anhydrous sodium sulfate. The solvent is distilled away under reduced pressure. The residue is purified by column chromatography on silica gel (eluent: hexane : ethyl acetate : triethylamine = 5 : 5 : 1) to obtain 2.2 g of the desired product as a colorless oily matter.

Cis form NMR (CDCl₃, δ , ppm): 2.24(s, 3H), 2.45(s, 8H), 2.94-3.32(m, 2H), 3.84(s, 3H), 5.22(bs, 2H), 5.85(t, 1H, J=6.8Hz), 6.66-8.07(m, 7H)

Mass spectrum (m/z): 378 (M⁺)

Trans form NMR (CDCl₃, δ , ppm): 2.24(s, 3H), 2.45(s, 8H), 2.94-3.32(m, 2H), 3.84(s, 3H), 5.22(bs, 2H), 6.22(t, 1H, J=6.8Hz)

Mass spectrum (m/z): 378 (M⁺)

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Example 23

Methyl 11-(2-morpholinoethylidene)-6,11-dihydro-dibenz[b,e]oxepin-2-carboxylate (Compound 12)

The desired product is obtained by substituting morpholine for 4-methylpiperazine in Example 22.

Example 24

Methyl 11-(2-thiomorpholinoethylidene)-6,11-dihydro-dibenz[b,e]oxepin-2-carboxylate (Compound 14)

The desired product is obtained by substituting thiomorpholine for 4-methylpiperazine in Example 22.

Example 25

Methyl 11-(2-pyrrolidinoethylidene)-6,ll-dihydro30 dibenz[b,e]oxepin-2-carboxylate (Compound 16)

The desired product is obtained by substituting

Example 26

Methyl ll-(2-piperidinoethylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 17)

pyrrolidine for 4-methylpiperazine in Example 22.

The desired product is obtained by substituting piperidine for 4-methylpiperazine in Example 22.

Example 27

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Methyl 11-[2-(4-methylpiperazino)ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-acetate (Compound 25)

The desired product is obtained by substituting methyl ll-methylene-6,ll-dihydrodibenz[b,e]oxepin-2-acetate for methyl ll-methylene-6,ll-dihydrodibenz[b,e] oxepin-2-carboxylate in Example 22.

Example 28

10 ll-(3-Dimethylaminopropylidene)-6,ll-dihydrodibenz [b,e]oxepin-2-carboxylic acid (Compound 3)

In this example, 26.1 g of methyl 11-(3-dimethyl-aminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxy-late is dissolved in a mixed solvent of 500 ml of methanol and 30 ml of water and 6.2 g of sodium hydroxide is added thereto. The mixture is heated at reflux for two hours. After allowing the mixture to stand for cooling, aqueous 4N-hydrochloric acid solution is added thereto to adjust the pH to 7 and the mixture is concentrated under reduced pressure. The concentrate is purified by column chromatography on high porous polymer (HP-20) (eluent: water: methanol = 1:2) to obtain 25.0 g of the desired product.

Cis form white crystal

Melting point: 162 - 164°C

25 NMR (DMSO-d₆, δ, ppm): 2.28(s, 6H), 2.40-2.70(m, 4H), 5.20-5.40(broad, 2H), 5.72(t, 1H, J=7.0Hz), 6.85-7.90(m, 7H)

IR (KBr disk, cm⁻¹): 3400, 1610, 1370, 1220, 1005 Elemental analysis (%): as $C_{20}H_{21}O_3N \cdot 1/3$ H_2O

C H N

Found: 73.00 6.67 4.14 Calculated: 72.93 6.63 4.25

Calculated: 72.93 6.63 4
Trans form white crystal

Melting point: 242 - 244°C

NMR (DMSO-d₆, δ, ppm): 2.25(s, 6H), 2.40-2.70(m, 4H), 5.20-5.40(broad, 2H), 6.09(t, lH, J=7.0Hz), 6.78-7.90(m, 7H)

IR (KBr disk, cm^{-1}): 3400, 1610, 1380, 1222, 1010 Elemental analysis (%):

C H N
Found: 74.30 6.60 4.30
Calculated: 74.28 6.55 4.30

Examples 29 - 34

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11-(3-Diethylaminopropylidene)-6,11-dihydrodibenz
[b,e]oxepin-2-carboxylic acid (Compound 5)

11-(3-Pyrrolidinopropylidene)-6,11-dihydrodibenz
[b,e]oxepin-2-carboxylic acid (Compound 7)

11-(4-Dimethylaminobutylidene)-6,11-dihydrodibenz
[b,e]oxepin-2-carboxylic acid (Compound 9)

ll-[2-(4-Methylpiperazino)ethylidene]-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic acid (Compound 11)

11-(2-Morpholinoethylidene)-6,11-dihydrodibenz[b,e]
oxepin-2-carboxylic acid (Compound 13)

11-(2-Thiomorpholinoethylidene)-6,11-dihydrodibenz
[b,e]oxepin-2-carboxylic acid (Compound 15)

These products are obtained by hydrolysis in the same manner as described in Example 28.

25	Compound	Melting point (°C)	Elementary analysis (%) or Mass spectrum		
30		White solid 120 - 123 (Acetonitrile)	Cis: Trans = 7: 3 As $C_{22}^{H_{25}O_3N}$ C H N Found 75.10 7.11 3.87 Calculated 75.19 7.17 3.99		
30	7	Colorless amor- phous solid About 150 (Decomposition)	For C ₂₂ H ₂₃ O ₃ N 349 (M ⁺)		

	Compound	Melting point (°C)	Elementary analysis (%) or Mass spectrum
5	9	White solid 128 - 129 (Water)	Cis:Trans = 9:1, dihydrate As C ₂₁ H ₂₃ NO ₃ ·2H ₂ O C H N Found 67.61 7.03 4.00 Calculated 67.54 7.29 3.75
10	11	White solid 150 - 153 (Water)	Cis:Trans = 1:9, dihydrate As C ₂₂ H ₂₄ NO ₃ ·2H ₂ O C H N Found 65.98 6.99 6.95 Calculated 65.98 7.05 7.00
15	13	White solid 130 - 133 (Toluene)	Cis: Trans = 1:9 As C ₂₁ H ₂₁ O ₄ N C H N Found 71.52 6.11 3.81 Calculated 71.78 6.02 3.99
20	15	Colorless amorphous solid About 140	As C ₂₁ H ₂₁ O ₃ NS 367 (M ⁺)

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Example 35

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11-(3-Dimethylaminopropylidene)-6,11-dihydrodibenz
[b,e]oxepin-2-acetic acid (Compound 20)

The product is obtained by hydrolysis as in the 30 same manner as in Example 28.

Cis form white crystal

Melting point: 118 - 120°C (Isopropanol)

NMR (DMSO-d₆, δ , ppm): 2.16(s, 6H), 2.30-2.60(m, 4H), 4.04(s, 2H), 5.15(bs, 2H), 5.69(t, 1H, J=7Hz),

6.73-7.40 (m, '7H)

IR (KBr disk, cm^{-1}): 3400, 1580, 1225, 1005 Mass spectrum (m/z): 337 (M⁺)

```
Elementary analysis (%):
                                       as C_{21}H_{23}O_3N · monohydrate
                                 С
                                         Η
                               70.77
                Found
                                        7.36
                                                3.74
                Calculated
                               70.96
                                        7.09
                                                3.94
          Trans form
                        white crystal
          Melting point:
                           158 - 160°C (Acetonitrile)
          NMR (DMSO-d_6, \delta, ppm): 2.05(s, 6H), 2.30-2.60(m, 4H),
                4.04(s, 2H), 5.15(bs, 2H), 6.06(t, 1H, J=7Hz),
                6.73-7.40 (m, 7H)
          IR (neat, cm^{-1}):
                              3380, 1575, 1220, 1005
10
          Mass spectrum (m/z): 337 (M^{+})
                                       as C_{21}H_{23}O_3N \cdot monohydrate
          Elementary analysis (%):
                               71.06
                Found
                                        6.66
                                                3.92
15
                               70.96
                                        7.09
                Calculated
                                                3.94
     Examples 36 - 39
          11-(4-Dimethylaminobutylidene)-6,11-dihydrodibenz
     [b,e]oxepin-2-acetic acid (Compound 22)
20
          11-(3-Pyrrolidinopropylidene)-6,11-dihydrodibenz
     [b,e]oxepin-2-acetic acid (Compound 24)
```

3-[11-(3-Dimethylaminopropylidene)-6,11-dihydrodibenz
[b,e]oxepin-2-yl]propionic acid (Compound 28)

dibenz[b,e]oxepin-2-acetic acid (Compound 26)

These products are obtained by hydrolysis in the same manner as in Example 35. The physicochemical properties are shown in Table 9.

11-[2-(4-Methylpiperazino)ethylidene]-6,11-dihydro-

Table 9

	Compound	Melting point (°C)	Elementary analysis (%)				
5	22	White solid 206 - 209	Cis : Trans = 92 : 8				
10		(Isopropanol)	Found Calculated	C 75.20 75.19			
	26	White solid 206 - 209 (Isopropanol)	Cis: Trans		н	И	
15			Found Calculated	*		1	

```
Compound 28
          Cis form
                      white crystal
          Melting point: 136 - 138°C (Isopropylether)
20
          NMR (DMSO-d_6, \delta, ppm): 2.32(m, 2H), 2.38(s, 6H),
               2.44-2.56(m, 2H), 2.73(m, 4H), 5.15(bs, 2H),
                5.50(m, 1H), 6.7-7.4(m, 7H)
          IR (KBr disk, cm^{-1}): 3380, 1645
          Mass spectrum (m/z): 351 (M^{+})
25
          Elementary anslysis (%):
                                      as C22H25NO3
                                 C
               Found
                              74.83
                                      .7.31
                                              3.97
                              75.19
                                       7.17
                                              3.99
               Calculated
30
          Trans form
                        white crystal
          Melting point: 148 - 149°C (Acetonitrile)
          NMR (DMSO-d_6, \delta, ppm): 2.05(s, 6H), 2.24(m, 2H),
                2.35(m, 2H), 2.47(t, 2H, J=7.5Hz), 2.72(t, 2H,
               J=7.5Hz), 4.80-5.50(broad, 2H), 5.99(t, 1H, J=
                7.1Hz), 6.6-7.5 (m, 7H)
35
          IR (KBr disk, cm^{-1}): 3380, 1700
```

Mass spectrum: 351 (M⁺)

```
as C_{22}^{H}_{25}^{NO}_{3} \cdot 1/5 hydrate
          Elementary analysis (%):
                             74.53
                                      7.20
               Found
                                             4.32
               Calculated
                             74.42
                                      7.21
                                             3.95
 5
     Example 40
          11-(2-Dimethylaminoethyl)imino-6,11-dihydrodibenz
     [b,e]oxepin-2-acetic acid (Compound 37)
                The desired product is obtained as a 8:92 mix-
10
     ture of cin-form and anti-form by hydrolysis in the same
     manner as in Example 27.
          White crystal
          Melting point:
                           174 - 176°C (as 1/2 hydrate)
          NMR (DMSO-d_6, \delta, ppm): 2.07(s, 6H), 2.30-2.80(m, 4H),
15
                3.47(s, 2H), 4.90-5.30(broad, 2H), 6.74-7.62
                (m, 7H)
          IR (KBr disk, cm^{-1}): 3350, 1575, 1370, 1010
                                      as C_{20}^{H}_{22}^{N}_{2}^{O}_{3} \cdot 1/2 hydrate
          Elementary analysis (%):
                                       Η
20
                Found
                              69.47
                                      6.77
                                              8.06
                Calculated
                              69.14
                                      6.67
                                              8.06
     Examples 41 - 47
          11-(2-Diethylaminoethyl)imino-6,ll-dihydrodibenz
25
     [b,e]oxepin-2-carboxylic acid (Compound 35)
          11-(2-Diethylaminoethyl)imino-6,ll-dihydrodibenz
     [b,e]oxepin-2-acetic acid (Compound 39)
          11-(3-Dimethylaminopropyl)imino-6,ll-dihydrodibenz
     [b,e]oxepin-2-acetic acid (Compound 41)
30
           3-[11-(2-Diethylaminoethyl)imino-6,11-dihydrodibenz
     [b,e]oxepin-2-yl]-propionic acid (Compound 43)
           2-[11-(2-Dimethylaminoethyl)imino-6,11-dihydrodibenz
     [b,e]oxepin-2-yl]-propionic acid (Compound 45)
           11-(2-Dimethylaminoethyl)imino-6,ll-dihydrodibenz
35
      [b,e]oxepin-3-acetic acid (Compound 47)
           11-(3-Dimethylaminopropyl)imino-6,ll-dihydrodibenz
     [b,e]oxepin-3-acetic acid (Compound 49)
```

The desired compounds are obtained by hydrolysis in the same manner as in Example 40. The physicochemical properties are shown in Table 10.

5

Table 10

	Compound	Melting point (°C)	Elementary or Mass spe		s (%)		
White solid Cin: Anti as $C_{21}^{H}_{24}^{O}_{3}$							
		(Isopropyl ether)	Found Calculated	C 71.66			
15	39	White solid	Anti: 98% as C ₂₂ H ₂₆ O ₃				
		(Ethyl acetate)	Found Calculated	C 72.25		l	
20	41	White solid 171 - 173	Anti: 97% as C ₂₁ H ₂₄ O ₃	N ₂		-	
		(Isopropanol)	Found Calculated	C 71.35 71.57		}	
25	43	Colorless Oil	as C ₂₃ H ₂₈ O ₃ 380 (M				
	45	White solid 132 - 135	Anti > 95% as C ₂₁ H ₂₄ O ₃	N ₂			
30		(Water)	Found Calculated	C 71.39		N 7.91 7.95	
•	47	White solid	Anti > 95% as C ₂₀ H ₂₂ O ₃ N ₂				
35		(Decomposition) (Methanol)	Found Calculated	C 70.87 70.98		N 7.93 8.28	

Compound	Melting point (°C)	Elementary analysis (%) or Mass spectrum				
49	White solid 174 - 175 (Decomposition) (Isopropanol)	Anti > 95% as C ₂₁ H ₂₄ O ₃ Found Calculated	N ₂ C 71.42	H 7.03 6.86	N 8.06	

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Example 48

Methyl 11-(3-dimethylaminopropyl)-6,11-dihydrodibenz [b,e]oxepin-2-carboxylate (Compound 50)

Process A:

11-Hydroxy-2-(4,4-dimethyl-2-oxazoline-2-yl)-6,11dihydrodibenz[b,e]oxepin

In this process, 2.40 g of ll-oxo-2-(4,4-dimethyl-2-oxazoline-2-yl)-6,ll-dihydrodibenz[b,e]oxepin is dissolved in 100 ml of methanol and 0.3 g of sodium borohydride is added thereto. After stirring the mixture at room temperature for 30 minutes, the solvent is distilled away under reduced pressure. The residue is extracted with 200 ml of methylene chloride, washed with saturated aqueous sodium bicarbonate solution and saturated aqueous sodium chloride solution in order, and dried over anhydrous sodium sulfate and the solvent is distilled away under reduced pressure. The residue is recrystallized from toluene to obtain 2.06 g of the desired product as a white solid.

Melting point: 201 - 203°C

Process B:

35 ll-(3-Dimethylaminopropyl)-2-[4,4-dimethyl-2-oxazo-line-2-yl)-6,ll-dihydrodibenz[b,e]oxepin

In this process, 1.90 g of ll-hydroxy-2-(4,4-dimethyl-2-oxazoline-2-yl)-6,ll-dihydrodibenz[b,e]oxepin is dissolved in 30 ml of methylene chloride and 0.7 ml of thionyl chloride is added thereto under ice-cooling.

After stirring the mixture at room temperature for one hour, the solvent is distilled away under reduced pressure to obtain a crude product of ll-chloro-2-(4,4-dimethyl-2-oxazoline-2-yl)-6,ll-dihydrodibenz[b,e]oxepin. The crude product is dissolved in 10 ml of tetrahydrofuran without purification.

To the solution is dropwise added under a nitrogen atmosphere 3-dimethylaminopropyl magnesium chloride obtained in the same manner as in Process C of Example 1 until the raw material is disappeared. The reaction mixture is extracted with 100 ml of methylene chloride, washed with saturated aqueous sodium bicarbonate solution and saturated aqueous sodium chloride solution in order and dried over anhydrous sodium sulfate, and the solvent is distilled away under reduced pressure. The residue is purified by column chromatography on silica gel (eluent: hexane: ethyl acetate: triethylamine = 10:10:1) to obtain 0.06 g of the desired product as a colorless oily matter.

Mass spectrum (m/z): 378 (M^+) for $C_{24}H_{30}O_2N$

Process C:

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Methyl 11-(3-dimethylaminopropyl)-6,11-dihydrodibenz-[b,e]oxepin-2-carboxylate

In this process, 60 mg of ll-(3-dimethylaminopropyl)-2-(4,4-dimethyl-2-oxazoline-2-yl)-6,ll-dihydrodibenz[b,e]oxepin is dissolved in a mixed solvent of 20
ml of water and 20 ml of dioxane and 10 mg of p-toluenesulfonic acid is added thereto. After heating the mixture
at reflux for 3 hours, the mixture is concentrated under
reduced pressure. The concentrate is extracted with 100
ml of ethyl acetate, washed with saturated aqueous sodium
bicarbonate solution and saturated aqueous sodium chloride

solution in order and dried over anhydrous sodium sulfate, and the solvent is distilled away under reduced pressure. The residue is dissolved in a mixed solution of 30 ml of methanol and 10 ml of aqueous lN-sodium hydroxide solution and the mixture is heated at reflux for 2 hours. After allowing the mixture to stand for cooling, the pH of the mixture is adjusted to 5.4 with aqueous 4N-hydrochloric acid solution.

The solvent is distilled away under reduced pressure and the residue is redissolved in 50 ml of methanol. After adding 10 mg of p-toluenesulfonic acid thereto, the mixture is heated at reflux for 3 hours and concentrated under reduced pressure. The residue is extracted with 100 ml of ethyl acetate, washed with saturated aqueous sodium bicarbonate solution and saturated aqueous sodium chloride solution in order and dried over anhydrous sodium sulfate and the solvent is distilled away under reduced pressure. The residue is developed on 3 sheets of preparative TLC (20 cm \times 20 cm \times 0.25 mm) with a mixed solvent (eluent: hexane : ethyl acetate : triethylamine = 10 : 10 : 2). The band at Rf = 0.47 is collected, and extracted with methylene chloride and the solvent is distilled away under reduced pressure to obtain 5.3 mg of the desired product as a colorless oily matter. NMR (CDCl₃, δ , ppm): 1.20-1.40(m, 1H), 1.60-1.80 (m, 2H), 2.18(m, 2H), 2.56(s, 6H), 2.74(dd, 2H,

J=6.6Hz and 9.5Hz), 3.90(s, 3H), 5.00 and 5.59 (ABq, 2H, J=14.2Hz), 6.96-7.88(m, 7H) Mass spectrum (m/z): 325 (M $^+$) for $C_{20}^{H}_{23}^{O}_{3}^{N}$ IR (neat, ν , cm $^{-1}$): 3400, 1710, 1610, 1110

Example 49

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1/2 Fumarate · 1/5 hydrate of Compound 3 (Compound 3')

In this example, 3.95 g of ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-carboxylic
acid (Compound 3) is dissolved in 100 ml of acetone and
1.42 g of fumaric acid is added thereto. The mixture is

stirred at room temperature. The deposited crystals are collected by filtration and recrystallized from isopropanol to obtain 4.15 g of the desired product as a white solid.

Melting point: 253 - 254°C

Isomer purity: Trans form 99% (measured by HPLC)

Elementary analysis (%):

as $C_{20}H_{21}NO_3 \cdot 1/2C_4H_4O_4 \cdot 1/5H_2O$

C H N

Found 68.74 6.35 3.61

Calculated 68.63 6.13 3.64

Examples 50 - 59

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The products identified in Table 11, the physicochemical properties of which are shown in Table 12 are obtained in the same manner as in Example 49.

Table 11

•			
20	Compound No.		
	5'	Monofumarate • 1/3 hydrate of Compound 5	(Cis form 99%)
25	7'	Monofumarate · monohydrate of Compound 7	(Cis form 70%)
	11'	Difumarate ·1/2 hydrate of Compound ll	(Trans form 100%)
	13'	1/2 Fumarate · 1/2 hydrate of Compound 13	(Trans form 93%)
30	15'	Monofumarate of Compound 15	(Trans form 100%)
	20'	Monofumarate · 3/2 hydrate of Compound 20	(Trans form 95%)
	26'	Monofumarate · 2/3 hydrate of Compound 26	(Trans form 88%)
35	28 ^{t.}	Monofumarate · 1/2 hydrate of Compound 28	(Trans form 63%)

Compound No.

31' 1/2 Fumarate · monohydrate of (Trans form 95%)
Compound 31

33' Monofumarate of Compound 33 (Cis form 100%)

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Table 12

	Compound	Melting point (°C)	Elementary analysis (%)			
15	5 '	White solid	as C ₂₆ H ₂₉ O ₇ N·1/3H ₂ O			
13		100	•	С	H	N
		(Decomposition)	Found	66.03	6.31	2.96
		(Isopropylether)	Calculated	66.14	6.55	3.14
•	7 '	White solid	as C ₂₆ H ₂₇ O ₇	N • Н ₂ О		,
20		vague owing to		C	H	N
		absorption of	Found	64.32	6.11	2.66
		moisture	Calculated	64.59	6.05	2.90
	11'	White solid	as C ₃₀ H ₃₂ O ₁	1 ^N 2·1/2	H ₂ O	
		266 - 268		C	Н	. И
25			Found	59.55	5.44	4.53
		(Isopropanol)	Calculated	59.50	5.49	4.63
	13'	White solid	as C ₂₃ H ₂₃ O ₆ N·1/2H ₂ O			
		232 - 235	. 23 20 0	С	Н	N
		(Decomposition)	Found	66.63	5.83	3.44
30		(Isopropanol)	Calculated	66.72	5.85	3.44.
	15'	White solid	as C ₂₅ H ₂₅ O ₇	NS	,	
		250 - 254			H	N
			Found	64.21	5.59	3.73
35		(Isopropanol)	Calculated	64.23	5.39	3.99
		<u></u>	<u> </u>			

	<u>·</u>					
	Compound	Melting point (°C)	Elementary	analysi	s (%)	
5	20'	White solid	as C ₂₅ H ₂₇ O ₇	N • 3/2H	2 ^О Н	N
		133 130	Found	62.58		2.77
		(Isopropyl ether)	Calculated			į
	26'	White solid	as C ₂₇ H ₃₀ O ₇	N ₂ .2/3H	20	
10		108 - 110	27 30 7	C	H	N
			Found	64.15	6.47	5.24
		(Isopropanol)	Calculated	64.02	6.24	5.53
	28'	White amorphous powder	as C _{2,6} H ₂₉ NO	•		·
15		vague owing to		C	Н	N
		absorption of	•	66.58		ľ
		moisture	Calculated	66.80	6.25	3.00
	31'	White solid	as C ₂₃ H ₂₇ O ₄	и • н ₂ о		:
;				С	H	N
20	-	vague owing to absorption of	Found	65.53	6.81	2.96
		moisture	Calculated	65.39	6.92	3.32
;		(Petroleum ether)			•	·
	33'	White solid	as C ₂₆ H ₃₁ O ₆	N		
25		146	20 32 0	С	Н	N
	,		Found _.	68.81	7.16	3.22
		(Acetone)	Calculated	68.86	6.89	3.09
			·			

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Example 60

Monosodium salt *monohydrate of Compound 35 (Compound 35')

In this example, 1.00 g of 11-(2-diethylamino-ethyl)imino-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid (Compound 35) is dissolved in 100 ml of methanol and

5.5 ml of 28% sodium methoxide methanol solution is added thereto. After stirring the mixture for one hour, the solvent is distilled away under reduced pressure. The residue is triturated with isopropylether and is collected by filtration to obtain 0.98 g of the desired product as a white solid.

Melting point: vague (hygroscopic)
Ratio of isomer Syn : Anti = 1 : 1
Elementary analysis: as $C_{21}^{H}_{25}^{O}_{4}^{N}_{2}^{Na} \cdot H_{2}^{O}$ C H N

Found 64.23 6.62 7.01
Calculated 64.27 6.68 7.14

Examples 61 and 62

The same procedures as in Example 60 are repeated to obtain the products identified in Table 13, the physicochemical properties of which are shown in Table 14.

20 <u>Table 13</u>

Compound No.		
43'	Sodium salt of Compound 43	(Anti form 98%)
45 '	Sodium salt · monohydrate of Compound 45	(Anti form 99%)

30

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Table 14

_	Compound No.	Melting point (°C)	Elementary	analysi	s (%)	
5	43'	White solid	as C ₂₃ H ₂₇ O ₃	N ₂ Na		·
		_		С	H	N ·
		vague owing to absorption of	Found	68.46	7.00	6.88
		moisture	Calculated	68.64	6.76	6.96
. 10	45'	White solid	as C ₂₁ H ₂₃ O ₃ N ₂ Na·H ₂ O			
		140 - 145		С	H	N
			Found	64.11	6.57	6.99
		(Isopropyl ether)	Calculated	64.27	6.42	7.14

15

Example 63 Tablet

A tablet comprising the following components is prepared in a conventional manner.

20 Trans-11-(3-dimethylaminopropylidene)-6,11dihydrodibenz[b,e]oxepin-2-carboxylic acid ·1/2 fumarate·1/5 hydrate (Compound 3'): 30 mg 60 mg Lactose: Potato starch: 30 mg 25 Polyvinyl alcohol: 2 mg Magnesium stearate: 1 mg q.s. Tar pigment:

30

35

Example 64 Powder

A powder comprising the following components is prepared in a conventional manner.

Trans-ll-(3-dimethylaminopropylidene)-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acidmonofumarate·3/2 hydrate (Compound 20'): 30 mg

Lactose: 270 mg

Example 65 Syrup

A syrup comprising the following components is prepared in a conventional manner.

11-(2-dimethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid

(Compound 37):

Purified sucrose: 40 g

Methyl p-oxybenzoate: 40 mg

300 mg

Propyl p-oxybenzoate: 10 mg

Strawberry flavor: 0.1 cc

Water is added to the above components until the total volume becomes 100 cc

Effect of the Invention

Compound (I) and a pharmaceutically acceptable salt thereof have an antiallergic and/or antiinflammatory activity.

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Applicant: (102) Kyowa Hakko Kogyo Co., Ltd.
Mikio Kato, Representative Director

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Example 65 Syrup

A syrup comprising the following components is prepared in a conventional manner.

11-(2-dimethylaminoethyl)imino-6,ll-dihydrodibenz[b,e]oxepin-2-acetic acid (Compound 37):

Purified sucrose: 40 g

300 mg

Methyl p-oxybenzoate: 40 mg

Propyl p-oxybenzoate: 10 mg

Strawberry flavor: 0.1 cc

Water is added to the above components until the total volume becomes 100 cc

15

10

Effect of the Invention

Compound (I) and a pharmaceutically acceptable salt thereof have an antiallergic and/or antiinflammatory activity.

20

Applicant: (102) Kyowa Hakko Kogyo Co., Ltd.
Mikio Kato, Representative Director

25